



**INSTITUTO POLITÉCNICO NACIONAL**



**Centro Interdisciplinario de Investigación  
para el Desarrollo Integral Regional Unidad Oaxaca**

**Doctorado en ciencias en conservación**

**y aprovechamiento de recursos naturales**

**(Patrones y procesos para la biodiversidad del neotrópico)**

***Estudio de señales ultrasónicas de materiales basados en  
cemento durante procesos de hidratación, carbonatación y  
penetración de cloruros***

**Tesis que para obtener el grado académico de:**

**Doctor en ciencias**

**Presenta:**

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**Dr. Francisco Castellanos León**

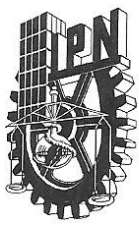
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# INSTITUTO POLITECNICO NACIONAL

## SECRETARIA DE INVESTIGACION Y POSGRADO

### ACTA DE REVISION DE TESIS

En la Ciudad de Oaxaca de Juárez siendo las 13:00 horas del día 29 del mes de noviembre del 2016 se reunieron los miembros de la Comisión Revisora de Tesis designada por el Colegio de Profesores de Estudios de Posgrado e Investigación del **Centro Interdisciplinario de Investigación para el Desarrollo Integral Regional, Unidad Oaxaca (CIIDIR-OAXACA)** para examinar la tesis de grado titulada: **“Estudio de señales ultrasónicas de materiales basados en cemento durante procesos de hidratación, carbonatación y penetración de cloruros**

Presentada por el alumno

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Apellido paterno

**López**

materno

**Mario Fernando**

nombre(s)

Con registro: 


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aspirante al grado de: **DOCTORADO EN CIENCIAS EN CONSERVACIÓN Y APROVECHAMIENTO DE RECURSOS NATURALES**

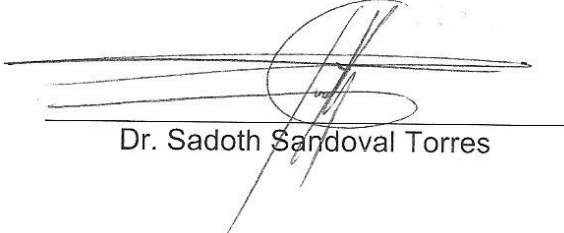
Después de intercambiar opiniones los miembros de la Comisión manifestaron **SU APROBACION DE LA TESIS**, en virtud de que satisface los requisitos señalados por las disposiciones reglamentarias vigentes.


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**CARTA CESION DE DERECHOS**

En la Ciudad de Oaxaca de Juárez el día 02 del mes diciembre del año 2016, el (la) que suscribe **Cosmes López Mario Fernando** alumno (a) del Programa de **DOCTORADO EN CIENCIAS EN CONSERVACIÓN Y APROVECHAMIENTO DE RECURSOS NATURALES** con número de registro B120151, adscrito al Centro Interdisciplinario de Investigación para el Desarrollo Integral Regional, Unidad Oaxaca, manifiesta que es autor (a) intelectual del presente trabajo de Tesis bajo la dirección de los Dres. Francisco Castellanos León y Prisciliano Felipe de Jesús Cano Barrita y cede los derechos del trabajo titulado: **“Estudio de señales ultrasónicas de materiales basados en cemento durante procesos de hidratación, carbonatación y penetración de cloruros”** Al Instituto Politécnico Nacional para su difusión, con fines académicos y de investigación.

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**Cosmes López Mario Fernando**



## Introducción general

El concreto es una pieza clave en la industria de la construcción en todo el mundo, como carreteras, presas, puentes, edificios, entre otros. Dichas estructuras están continuamente sujetas a efectos que degradan su integridad estructural provocando un riesgo de seguridad, por lo tanto, se deben desarrollar alternativas para una evaluación efectiva de su calidad sin la necesidad de dañar las estructuras estudiadas [1].

En este sentido, el uso de pruebas no destructivas constituye una interesante estrategia para el monitoreo y evaluación del estado de las estructuras de concreto sin dañar su apariencia o rendimiento [2]. Las pruebas no destructivas con ultrasonido se basan en la propagación de ondas a través del elemento analizado y han sido ampliamente aplicadas en estructuras de concreto [3,4]. La velocidad de propagación de las ondas ultrasónicas depende de distintas características del material como: densidad, módulo de elasticidad ( $E$ ) y coeficiente de Poisson, ( $\mu$ ); sus principales aplicaciones son la detección de daños y evaluación de condiciones iniciales del concreto [5].

En general la aplicación de los análisis no destructivos por ultrasonido se ha hecho usando dos métodos convencionales, la transmisión directa y el pulso eco [6]. En el método de transmisión se realiza la medición en lados opuestos de la estructura, para conocer el tiempo de recorrido que junto con la distancia recorrida nos permite conocer la Velocidad de Pulso Ultrasónico (UPV). El pulso eco, mide la reflexión por medio del mismo u otros transductores que actúan como receptores del pulso ultrasónico. Las pruebas anteriores se han estandarizado por medio de

procedimientos, instrumentaciones y análisis de datos para distintos tipos de ondas aplicadas a estructuras de concreto [7].

Cualquier variación en la velocidad de propagación de las ondas ultrasónicas indican vacíos, no uniformidad del material o daños [8,9]. Como señalan varios autores [10,11,12], las ondas ultrasónicas pueden verse afectados por varios factores, como la condición superficial, la humedad, la presencia de refuerzos y principalmente la composición del concreto. Tales factores provocan variaciones no solo en la velocidad de propagación de las ondas ultrasónicas, sino también en la señal en el tiempo, lo que dificulta el análisis de resultados [13]. Sin embargo, este hecho abre una ventana de oportunidad para poder conocer otras propiedades además del VPU, que permitan la evaluación de distintos fenómenos de una manera completamente no destructiva.

Debido a que la mayor cantidad de reacciones que suceden dentro del concreto se llevan a cabo en la pasta de cemento, el siguiente trabajo se centra en el análisis del comportamiento de la pasta de cemento en distintas condiciones, por medio de ondas longitudinales y transversales, con un intervalo de frecuencia de 50 kHz a 2250 kHz. Este trabajo se encuentra dividido en tres partes:

La primera parte, presenta el estudio de especímenes de concreto con el objetivo de identificar frecuencias específicas relacionadas con agregados y pasta de cemento durante la hidratación del concreto, por medio del análisis de transformada de Fourier de la respuesta ultrasónica de especímenes de concreto con diferentes frecuencias de excitación. Con lo que fue posible identificar el comportamiento de la pasta de cemento y agregados en el dominio de la frecuencia.



En la segunda parte, se estudió la respuesta ultrasónica en pastas de cemento con tres relaciones agua cemento ( $a/c = 0.60, 0.50$  y  $0.40$ ). Los especímenes fueron expuestos a la carbonatación acelerada en una cámara ambiental a  $30\text{ }^{\circ}\text{C}$ ,  $65\%$  de humedad relativa, y  $4\%$   $\text{CO}_2$  (v / v). Los resultados muestran que es posible evaluar el grado de carbonatación de los especímenes de pasta de cemento por medio del análisis en el dominio de frecuencia en función del tiempo de exposición, de los cambios relacionados a la carbonatación, independientemente de la relación agua / cemento.

La tercera parte presenta, presenta el estudio de dieciocho especímenes de pasta de cemento con una relación  $a / c = 0.55$ , con  $100\%$  de cemento Portland ordinario, sustitución del  $20\%$  y  $40\%$  de cemento por cenizas volantes de clase F y sustitución al  $10\%$  por humo de sílice. Los especímenes fueron secados en horno y saturados a  $4,08$  bar con soluciones de NaCl de  $16,5\%$  ( $2,8$  M),  $33\%$  ( $5,6$  M) y una con agua desionizada, como control. Los resultados muestran que, aplicando la metodología utilizada en las secciones anteriores y tomando como base los resultados del control, es posible distinguir entre las dos concentraciones de NaCl independientemente de las adiciones minerales y entre las pastas de cemento con adiciones minerales dependiendo de su capacidad de ligado.

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1 ***Capítulo I***

2 Ultrasound frequency analysis for identification of aggregates and cement  
3 paste in concrete

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14 **Abstract**

15 The aim of the present study was to identify specific frequencies related to aggregates  
16 and cement paste during concrete hydration, by performing a Fourier analysis of the  
17 ultrasonic response of concrete specimens to different excitation frequencies. This  
18 identification will reduce the high influence of aggregates in the ultrasound signal  
19 analysis, enabling a better assessment of changes occurring in the cement paste. Thirty-  
20 five cylindrical specimens with a diameter of 100 mm and a length of 200 mm were cast  
21 with a water to cement ratio = 0.60. Thirty specimens were destructively tested at 1, 3,  
22 5, 7, 14 and 28 days for their compressive strength. The remaining five specimens were  
23 non-destructively tested at 1, 3, 5, 7, 14, 28 and 56 days using longitudinal and  
24 transversal ultrasonic wave transducers with frequencies from 50 kHz to 500 kHz.  
25 Analysis of the evolution in frequency observed in the specimens identified variations  
26 related to progressive hydration of the cement paste, in contrast with the invariant  
27 behavior of the inert aggregates. Results show that it is possible to distinguish the  
28 behavior of cement paste and aggregates in the frequency domain. As a consequence, it  
29 should be possible in future research to evaluate more efficiently different phenomena  
30 that affect only the cement paste.

31 **Keywords**

32 Ultrasound; Fourier analysis; frequency bands; cement paste; aggregates; hydration

33

## 34 **1. Introduction**

35 Non-destructive techniques are used to evaluate the quality and performance of concrete  
36 structures without causing any structural damage [1, 2]. The Ultrasonic Pulse Velocity  
37 (UPV) is one of the most used non-destructive techniques for measuring mechanical  
38 properties of concrete during hydration and evaluating damage in concrete structures [3-  
39 6]. Previous studies have found an exponential relationship between the compressive  
40 strength of concrete and the UPV, which has shown to be barely sensitive to small  
41 variations in strength at late ages [7-8]. Concrete is basically composed of a mixture of  
42 cement, water, fine and coarse aggregates that produce a heterogeneous material. The  
43 heterogeneities produced by the coarse aggregates strongly influence the propagation of  
44 ultrasonic waves, depending on the frequency studied [9]. Therefore, the analysis of  
45 changes in an ultrasound wave propagated across a concrete specimen at selected  
46 frequencies, may be used to extract information on processes occurring mainly in the  
47 cement paste, for example during hydration or during chemical reactions in the cement  
48 paste with aggressive species (e.g. chlorides and CO<sub>2</sub>).

49 When a wave propagates, its phase and amplitude are modified depending on the  
50 transmission medium. The propagation of ultrasound waves is affected by the  
51 heterogeneities of concrete in different ways. The influence depends on the wavelength  
52 and the size of the concrete components [10]. In previous studies, this behavior was  
53 used to characterize entrained air voids in cement paste [11]. Other studies have  
54 employed the propagation velocity of the wave attenuation to estimate cracking [12-14],  
55 voids, the setting process [15, 16] and the compressive strength of concrete. Taking  
56 advantage of these properties, several experimental studies using through-transmission  
57 ultrasonic measurements of longitudinal or transversal waves have been carried out on

58 concrete to determine the setting time of concrete [4,17]. It has also been found that the  
59 attenuation of the ultrasound signal is different for cement paste, mortar and concrete in  
60 a frequency range of 50 - 1000 kHz [18-19]. Additionally, the setting process of cement  
61 pastes has been analyzed by the response of ultrasonic waves using the frequency  
62 spectrum of the Fast Fourier Transform [20].

63 Based on the aforementioned, it is assumed that, depending on the frequency band used,  
64 the response of concrete materials to ultrasonic waves contains more information than is  
65 reported. Therefore, this study focuses on developing a method for the identification of  
66 frequency bands associated with aggregate and cement paste during the hydration  
67 process of concrete. This identification would enhance the study of process which occur  
68 only in the cement paste such as hydration, chloride penetration and binding as well as  
69 carbonation produced by the chemical reaction between the  $\text{CO}_2$  and  $\text{Ca}(\text{OH})_2$  dissolved  
70 in the pore fluid.

## 71 **2. Experimental procedure**

### 72 **2.1. Materials**

73 Mexican ordinary Portland cement class 30RS (ASTM C150 Type I or EN 197-I, with  
74 chemical composition shown in Table 1) was used. The coarse aggregate was limestone  
75 with a maximum size of 13 mm. The fine aggregate was river sand with a fineness  
76 module of 2.67. Thirty-five cylindrical specimens measuring 100 mm in diameter and  
77 200 mm in height were cast with a w/c ratio of 0.60. After 1 day, the specimens were  
78 stripped and moist-cured in a saturated lime solution at  $23 \pm 2$  °C for 56 days, according  
79 to the standard ASTM C31 [21]. Moist curing with saturated lime solution avoids

80 leaching of the calcium hydroxide from the cement paste, which otherwise would  
81 increase the porosity and reduce the density and elastic modulus of the cement paste.

## 82 **2.2. Methods**

### 83 **2.2.1. Compressive strength**

84 Thirty specimens were tested at 1, 3, 5, 7, 14, and 28 days to measure their compressive  
85 strength using a 120 ton compression machine.

### 86 **2.2.2. Ultrasonic measurements**

87 Five specimens were tested at 1, 3, 5, 7, 14, 28 and 56 days. The ends of the specimens  
88 were cut with a concrete saw in order to obtain a smooth surface. A record was  
89 maintained of their responses in through-transmission to longitudinal and transversal  
90 ultrasonic waves, generated by eight pairs of transducers with frequencies ranging from  
91 50 kHz to 500 kHz. The position of the transducers and their applied pressure were kept  
92 constant during the experiment. Petroleum jelly was used as coupling agent to improve  
93 wave transmission. The excitation signal was generated by a 5058PR pulser/receiver  
94 with a voltage of 200 V and damping of 40 dB. The UPV was also obtained for the  
95 thirty specimens before compressive strength testing.

## 96 **3. Wave analysis**

### 97 **3.1. Signal processing**

98 The aim of the proposed method for analyzing the ultrasound signals is to identify  
99 frequency bands over time associated with aggregates and cement paste in hydraulic  
100 concrete. One of the features of concrete is that the properties of the hydrating cement



101 paste vary over time as opposed to those of aggregates which are invariant over time.  
 102 The Fast Fourier Transform (FFT) has been recognized as a standard signal processing  
 103 technique for analyzing ultrasound signals [22]. The first step in the proposed analysis  
 104 is to calculate the time-discrete Fourier Transform (TDFT) of the ultrasound signal  
 105  $f(n\Delta t)$ , according to Eq. (1).

$$106 \quad F(k\Delta\omega) = \frac{1}{2\pi} \sum_n f(n\Delta t) e^{-i(2\pi kn/N)\Delta t} \quad (1)$$

107 where  $T$  is the duration of the ultrasound signal, discretized at a sampling rate  $\Delta t = T/N$ ,  
 108 and  $k\Delta\omega$  are the resulting discrete frequencies.

109 A window of  $F(k\Delta\omega)$  is subsequently selected from the frequency response function of  
 110 the transducers with an intensity greater or equal to 6 dB, as specified by the ultrasound  
 111 transducer manufacturer.

$$112 \quad G(k\Delta\omega) = \begin{cases} F(k\Delta\omega), & k_1 \leq k \leq k_2 \\ 0, & elsewhere \end{cases} \quad (2)$$

113 The evolution in time of the seven spectra obtained (1, 3, 5, 7, 14, 28 and 56 days) from  
 114 ultrasound measurements  $G_i(k\Delta\omega)$ , where  $1 \leq i \leq 7$ , was analyzed for the  
 115 identification of frequency bands associated with aggregates or cement pastes. This  
 116 analysis consists of the correlation of variations along time of every frequency  
 117 amplitude  $|G_i(k\Delta\omega)|$ , with a function that describes an assumed behavior of aggregates  
 118 or cement pastes. For the identification of frequencies  $k_a$  associated with aggregates, a  
 119 correlation with a constant function different from zero was performed. In the case of  
 120 frequencies  $k_p$  related to cement pastes, they were correlated with the evolution in time  
 121 of the compressive strength that indicates the progress of cement hydration, Fig. 1. The

122 compressive strength was determined according to the standard ASTM C39 / C39M –  
123 16 [23]. The load was applied at a rate of  $0.25 \pm 0.05$  MPa/s ( $35 \pm 7$  psi/s).

124 Evolution of the compressive strength of concrete is closely related to density ( $\rho$ )  
125 changes in the hydrating cement paste, influencing the propagation of the ultrasonic  
126 waves [2]. Constant density of limestone aggregate ( $\rho = 2.60$  kg/m<sup>3</sup>), of sand ( $\rho = 2.57$   
127 kg/m<sup>3</sup>) and of saturated concrete were obtained, Fig. 1. A strong correlation coefficient  
128 ( $R=0.98$ ) states the evident similarity between density and compressive strength  
129 evolution in time. Similarly, a strong correlation ( $R=0.984$ ) was found between the  
130 compressive strength of concrete ( $f^c$ ) and UPV.

131 Once the assumed behavior for cement paste was identified, a threshold was defined as  
132 the mean value of the correlations obtained, with a precision of 0.001, for the  
133 identification of cement paste or aggregates, assuming a normal distribution for those  
134 values. The frequencies with correlations stronger than the threshold were assumed to  
135 be related to the corresponding behavior of cement paste or aggregates. As the  
136 thresholds for both identifications were independently established, there were atypical  
137 frequencies identified with both behaviors that were excluded from the analysis. Finally,  
138 the correlations between the energy ( $E_i$ ) and the functions that describe the assumed  
139 behavior were calculated

$$140 \quad E_i = \sum_{\forall j} |G_i(k_j \Delta \omega)|^2 \quad (3)$$

141 where  $j$  denotes the frequencies associated with paste or aggregates, respectively. The  
142 time evolution of the calculated energy must be consistent with the corresponding  
143 assumed behavior. The wave attenuation and wavelength analysis are two alternative

144 validations of the proposed method for identifying cement paste and aggregates,  
145 presented in the following sections.

### 146 **3.2. Wave attenuation analysis**

147 The different components in concrete induce a high attenuation mainly correlated with  
148 the size of aggregates. This attenuation  $\alpha$  is defined as [24]

$$149 \quad \alpha = -\frac{20}{x} \log \left( \frac{A_x}{A_0} \right) \quad (4)$$

150 where  $A_0$  is the initial amplitude of the wave and  $A_x$  is the amplitude after the wave has  
151 traveled across the length  $x$ .

152 By using Eq. (4), different attenuation responses were related to cement paste, mortar  
153 and concrete [19]. These different behaviors can be utilized to distinguish between  
154 concrete and cement paste depending on the frequency used. The attenuation can be  
155 correlated with the scattering of ultrasonic waves through the material, and in the case  
156 of concrete, this scattering has been linked to the presence of aggregates [19].  
157 Philippidis and Aggelis [19] found that the attenuation in cement paste specimens is  
158 higher in low frequencies than in high frequencies relative to the attenuation obtained in  
159 concrete specimens. Other studies support these findings [18, 25]. If the identified  
160 frequency bands are representative of the hardening process of the cement paste or the  
161 time invariant behavior of aggregates, then a comparison between the total energy of the  
162 signal and the energy of cement paste and aggregates must present a similar behavior of  
163 attenuation described in previous works [18, 19, 25]. This comparison was made  
164 possible by modifying Eq. (4) as follows.

165  $\alpha_{CP} = -\frac{20}{185} \log \left( \frac{E_{CP}}{E_o} \right)$  (5)

166  $\alpha_A = -\frac{20}{185} \log \left( \frac{E_A}{E_o} \right)$  (6)

167 where  $\alpha_{CP}$  and  $\alpha_A$  are the cement paste and aggregates attenuation, respectively.  $E_{CP}$   
 168 and  $E_A$  is the energy of the cement paste and aggregate frequency bands, respectively.  
 169  $E_o$  is the total energy of the signal and the mean length of the specimens were 185 mm.

170 **3.3. Wavelength analysis**

171 The wavelength  $\lambda$  depends on wave velocity ( $UPV$ ) and transducer frequency ( $\Omega$ )  
 172 according to

173  $\lambda = \frac{UPV}{\Omega}$  (7)

174 The size of discontinuities in concrete detected by ultrasound signals depends on its  
 175 wavelength [26]. The sensitivity for detecting a discontinuity of a given size increases  
 176 with higher frequencies as the wavelength decreases, assuming constant  $UPV$ . Concrete  
 177 contains approximately 65 – 70 % of aggregates and 30 – 35% of cement paste by  
 178 volume. This generates a thin and continuous layer of cement paste covering and filling  
 179 the empty spaces among aggregates. Therefore, the wavelengths produced by low  
 180 frequency transducers are more sensitive for detecting aggregates than cement paste, in  
 181 contrast to high frequency transducers, which preferentially detect cement paste.

182 For the case of concrete, the ratio of transducer diameter ( $D$ ) to the wavelength of the  
 183 transmitted wave ( $\lambda$ ) can be used to determine the radiation pattern ( $\beta$ ) of the stress  
 184 pulse in concrete, according to [27],

185 
$$\beta = \frac{D}{\lambda} \tag{8}$$

186 Most of the energy transmitted into the test specimen is contained within a cone-shaped  
187 region that has its base at the transducer surface and its highest amplitude in the normal  
188 direction of the transducer surface. The radiation pattern for each transducer, as well as  
189 its diameter are shown in Table 2. The radiation pattern mainly influences the wave  
190 reflection from the boundaries of the specimen. In our experiments, as the  $\beta$  index  
191 increases, it is expected that the wave reflected from boundaries of the specimen would  
192 decrease, as well as its influence on the received signal. This influence diminishes even  
193 more for higher frequencies due to the high attenuation of concrete [19]. However,  
194 further studies on the effect of radiation pattern on our experiments, under similar  
195 conditions, are necessary.

## 196 **4. Results and discussion**

### 197 **4.1. Frequency-domain identification of aggregates and cement paste**

198 Some typical results on the identification of frequencies related to cement paste and  
199 aggregates are shown in Figs. 2 to 5. The rest of the figures are found in appendix A.  
200 The frequencies associated with cement paste or aggregates are those with amplitudes  
201 higher than zero in the corresponding testing age-frequency plots. The close proximity  
202 of characteristic frequency bands of cement paste and aggregates can be explained by  
203 similar values of the modulus of elasticity of limestone aggregates (10.8 GPa) [28] and  
204 cement paste with a w/c = 0.60 (13.38 GPa) [29]. This property has a significant  
205 influence on the ultrasonic response of any material, especially for those subjected to  
206 longitudinal waves. The response obtained from transducers with the same frequency

207 but with different types of wave, show that the response of a transversal wave has a  
208 better definition of the frequency bands related to cement paste and aggregates, as can  
209 be seen by comparing Figs. 2 to 5.

#### 210 **4.2. Estimation of energy evolution over time of selected frequency bands for** 211 **aggregates and cement paste**

212 It is expected that the energy associated with aggregates remains constant in time, while  
213 the energy related to cement paste is assumed to increase, according to the changes in its  
214 compressive strength or density. Fig. 6 (transversal wave transducers) and Fig. 7  
215 (longitudinal wave transducers) present the evolution in time of energies produced by  
216 different frequency transducers and types of waves. The energies are calculated for the  
217 identified frequency bands and are related to the cement paste and to the aggregates. In  
218 Fig. 6a, it is shown that for the 50 kHz transducers, the energy associated with  
219 aggregates is higher than that associated with cement paste, in contrast to the other  
220 frequencies shown in the same figure. There it is shown that, at times no later than 14  
221 days after casting, the energy of cement paste reaches higher values than those values  
222 corresponding to aggregates. This behavior may be attributed to the higher value of the  
223 wavelength (53 mm at early stages) of the 50 kHz transversal transducer in comparison  
224 with the rest of the transducers. As a consequence, the ultrasound waves collect  
225 information from larger particles such as aggregates (maximum aggregate size of 13  
226 mm) rather than cement paste in the specimen. This phenomenon is accentuated because  
227 concrete is constituted by approximately 65-70% aggregates and 35-30% cement paste  
228 by volume, and as wavelength increases with cement paste hardening.

229 Fig. 7 shows the results for the longitudinal wave transducers. It is observed that the  
230 energy associated with cement paste is always higher than the energy of the aggregates  
231 after three days of age. This behavior can be attributed to the higher sensitivity of  
232 longitudinal waves to changes in density during concrete hardening, in comparison with  
233 transversal transducers. This is due to the fact that longitudinal waves which propagate  
234 in fluids and solids, have a greater interaction with the cement paste hardening in time,  
235 which is reflected in the corresponding energy increase.

### 236 **4.3. Wave attenuation**

237 The analysis of the attenuation was performed to increase the certainty that the energy  
238 bands are in fact related to the hardening process of cement paste and to aggregates.

239 The attenuation obtained from the energy associated with the frequency bands which  
240 were identified as cement paste or aggregates (Fig. 8a) is also consistent with results  
241 obtained by previous research [18, 19, 25]. It was found a direct correlation between the  
242 addition of aggregates and a higher attenuation of concrete in high frequencies relative  
243 to that of cement paste, as opposed to the higher attenuation associated with cement  
244 paste in low frequencies. For the longitudinal wave transducers, the resulting  
245 attenuation is similar to that obtained for transversal waves in the range of studied  
246 frequencies, where the aggregates attenuation is always higher relative to the cement  
247 paste attenuation (Fig. 8b).

248 The attenuation obtained for transversal wave transducers is shown in Fig. 8a, which  
249 yielded results as expected [18, 19, 25] with a higher attenuation of cement paste energy  
250 than aggregates energy in low frequencies; however, as the frequencies increase, they  
251 have a similar behavior until the attenuation of aggregates energy is higher in

252 comparison with cement paste energy at frequencies corresponding to 500 kHz  
253 transversal wave transducer.

254 In the case of longitudinal wave transducers (Fig. 8b), tested only with high frequencies,  
255 an evident attenuation difference between cement paste and aggregates can be  
256 appreciated, showing the higher attenuation of the aggregates energy bands in  
257 comparison with the cement paste energy bands in every transducer employed.

258 The average attenuation associated with the hardening of the cement paste for the  
259 longitudinal wave transducers is lower than that obtained for the transversal wave  
260 transducers. As the energy of frequency bands was used for the estimation of  
261 attenuation, due to the propagation properties of the wave, this behavior highlights the  
262 higher energy of the cement paste frequency bands obtained with the longitudinal wave  
263 transducer. In the case of the aggregates, the average attenuation obtained in 100, 250  
264 and 500 kHz is similar for both wave type transducers.

#### 265 **4.4. Wavelength analysis**

266 Wavelength depends on the frequency and the velocity of the wave Eq. (7). In turn,  
267 wave velocity propagation is a function of the properties of the materials such as density  
268 ( $\rho$ ), elastic modulus for longitudinal and shear modulus for transversal waves related to  
269 the concrete hardening [30]. These changes in the material are described by different  
270 wavelengths which influence the ability to differentiate between cement paste hardening  
271 and aggregates [31, 32]. The wavelengths obtained for this experiment are shown in  
272 Table 2.



273 Variations in wavelength for a given frequency, Table 2, are caused by the evolution of  
274 the hardening cement paste in concrete. These wavelength values are consistent with the  
275 energy behavior presented in Fig. 6 and Fig. 7.

276 In the case of the 50 kHz transversal wave transducers, produced a higher energy  
277 associated with aggregates, than that produced with cement paste (Fig. 6a). This is  
278 because of the wavelengths associated with this frequency ( $56.9 \pm 0.22$  mm) which have  
279 a higher interaction with aggregates (maximum size of 13 mm) than with cement paste,  
280 in addition to a higher volume percentage of aggregates in concrete (70%). As a  
281 consequence, the attenuation related to the cement paste for this low frequency  
282 transducer is high Fig. 8a.

283 In contrast, the 500 kHz transversal wave transducers, with wavelengths of  $3.6 \pm 0.02$   
284 mm, produce a higher energy associated with cement paste than with aggregates (Fig.  
285 6e) resulting in an opposite attenuation behavior. The rest of the frequencies (Figs. 6b -  
286 6d) present a transitional behavior between these two extremes. The resulting  
287 attenuation behaviors (Fig. 8a) are consistent with previous works [18, 19, 25].  
288 Additionally, the sensitivity of transversal wave transducers can be considered to remain  
289 constant through evaluation days as shown in the corresponding standard deviation  
290 (Table 2).

291 Regarding the longitudinal waves, the increase in wavelength in comparison with  
292 transversal waves, is directly proportional to their higher velocity [33] assuming a  
293 constant frequency Eq. (7). This increase is more remarkable at late ages in 250 kHz  
294 and 500 kHz frequency transducers and proportional to their standard deviations (Table  
295 2). The constant sensitivity of transversal wave transducer allows a better evaluation of

296 the hardening process of the cement paste at late ages in contrast to the sensitivity of  
297 longitudinal wave which decreases in time proportional to their standard deviations,  
298 especially at high frequencies such as 500 kHz.

299 The wavelength of the longitudinal wave transducers employed (100 – 500 kHz) is  
300 similar to those wavelengths corresponding to transitional attenuation behavior obtained  
301 for 100 – 250 kHz transversal wave transducers (Table 2 and Fig. 8a). Therefore, the  
302 attenuation behavior of longitudinal waves is inconsistent with previous works [18, 19,  
303 25], however if higher and lower frequencies are employed, the attenuation behavior is  
304 expected to be consistent.

## 305 **5. Conclusions**

306 Analysis of the results in the present paper allows for the following conclusions:

- 307 (1) The proposed signal analysis procedure allows for the identification of  
308 frequency bands related to cement paste and aggregates, whose assumed  
309 behavior in time is depicted by the evolution of compressive strength and a  
310 constant function, respectively. The attenuation analysis of frequency bands  
311 associated with cement paste and aggregates is consistent with previous  
312 research.
- 313 (2) In this study, due to their high correlation, UPV might be used to describe the  
314 behavior associated with cement paste instead of the actual compressive  
315 strength.
- 316 (3) Transversal wave transducers (especially at 250 kHz) allow for a better  
317 definition of the frequency bands associated with the cement paste and

318 aggregates. In addition, there is a higher correlation with attenuation in  
319 comparison with the longitudinal wave transducers.

320 (4) During the hydration of concrete, the smaller variations in the wavelength of  
321 transversal waves provide a higher sensitivity for detecting frequency bands  
322 associated with cement paste and aggregates in comparison with those  
323 obtained for longitudinal waves.

### 324 **Future work**

325 Research is under way aimed to improve non-destructive assessment of compressive  
326 strength by applying the method outlined in this paper. It will also be applied for  
327 detecting chemical reactions that only occur in the cement paste caused by aggressive  
328 species such as chloride ions and carbon dioxide.

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435 **Figure captions**

436 Fig 1. Evolution of concrete compressive strength (blue triangle) and saturated density  
437 (green squares) versus age of concrete  $w/c = 0.60$ .

438 Fig. 2. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
439 transducers used were 100 kHz longitudinal wave.

440 Fig. 3. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
441 transducers used were 100 kHz transversal wave.

442 Fig. 4. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
443 transducers used were 250 kHz longitudinal wave.

444 Fig. 5. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
445 transducers used were 250 kHz transversal wave.

446 Fig. 6. Evolution in time of energies associated with cement paste and aggregates versus  
447 testing age for transversal wave transducers (a) 50 kHz, (b) 100 kHz, (c) 180 kHz, (d)  
448 250 kHz and (e) 500 kHz

449 Fig. 7. Evolution in time of energies associated with cement paste and aggregates versus  
450 testing age for longitudinal wave transducers (a) 100 kHz, (b) 250 kHz and (c) 500 kHz

451 Fig. 8. Attenuation related to cement paste and aggregates energy bands, for (a)  
452 transversal and (b) longitudinal wave transducers

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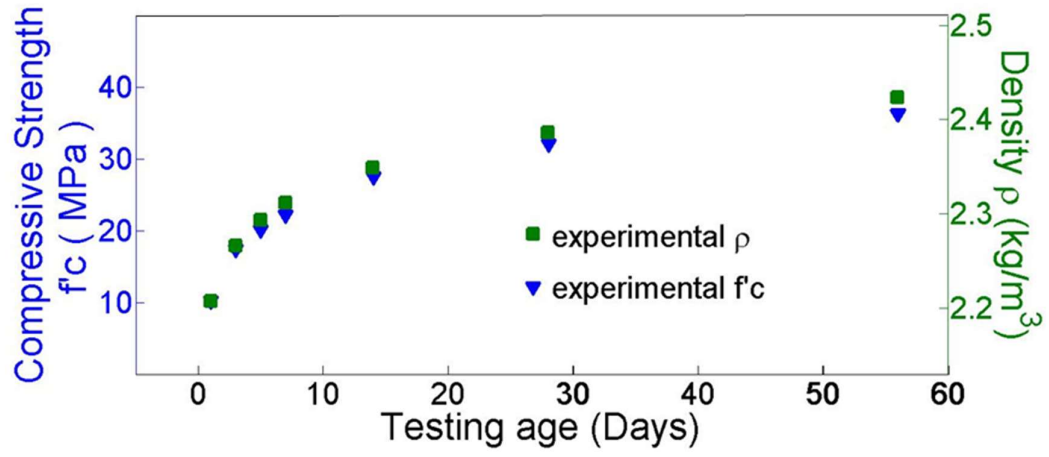


Fig 1.

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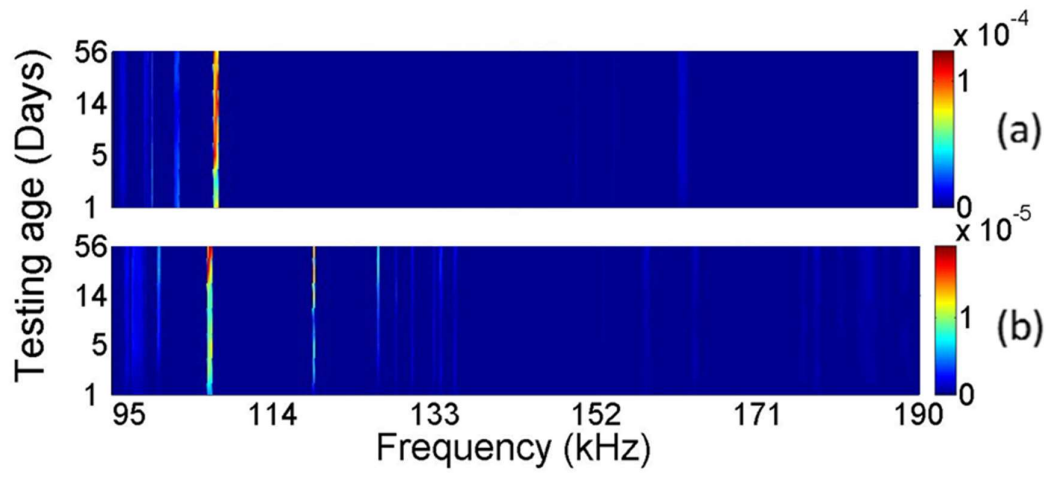


Fig. 2.

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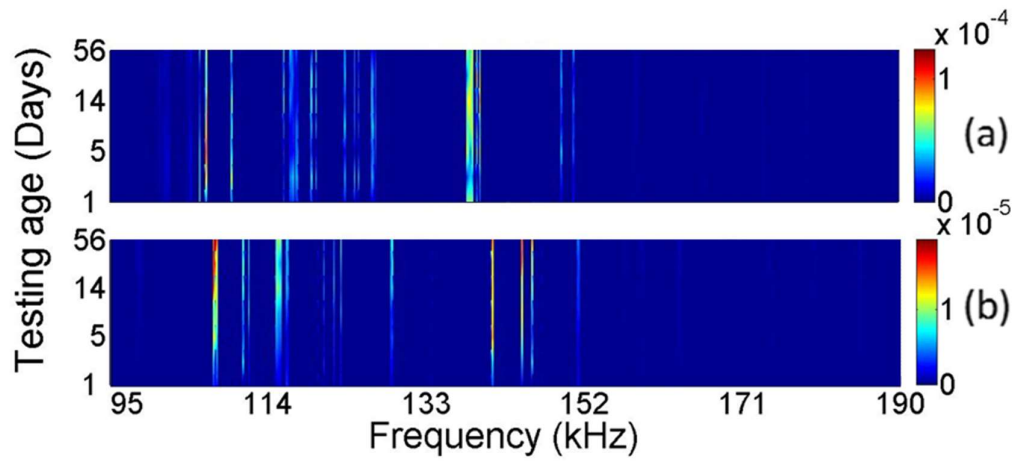


Fig. 3.

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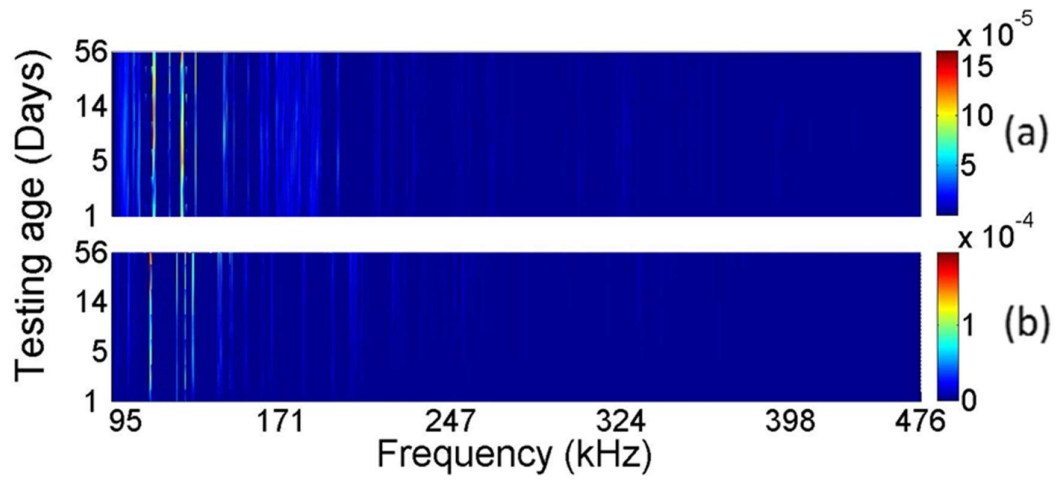


Fig. 4.

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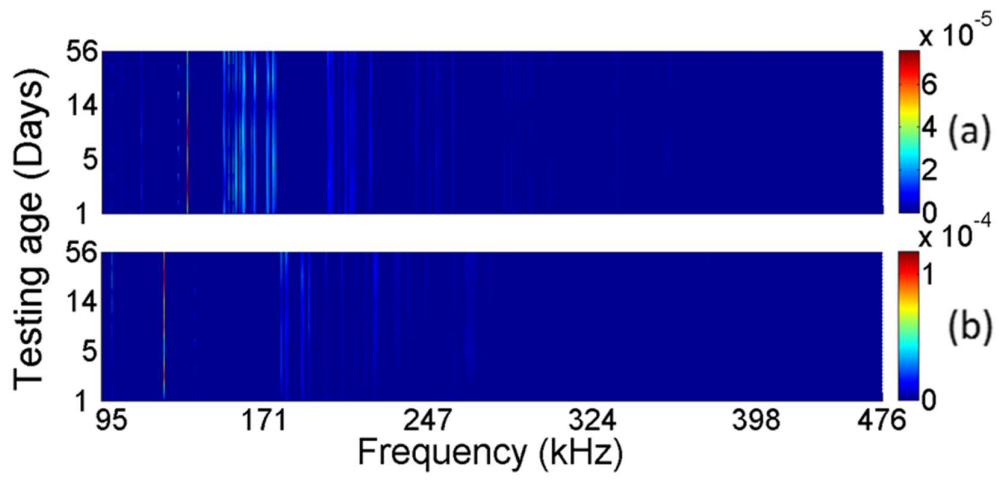


Fig. 5.

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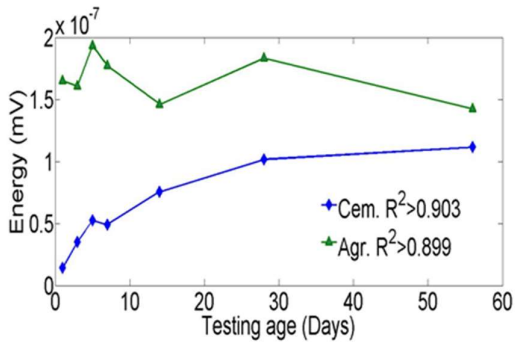
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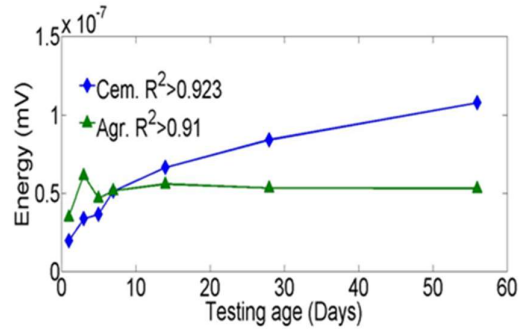
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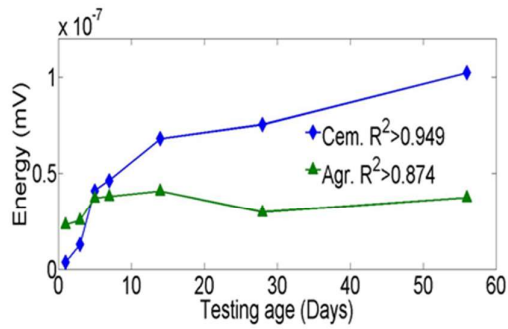
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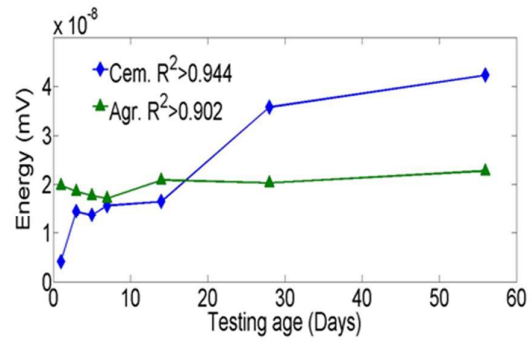
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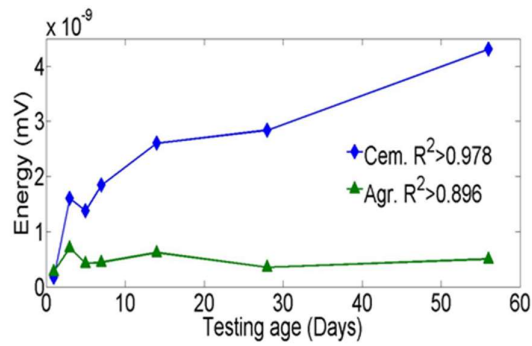
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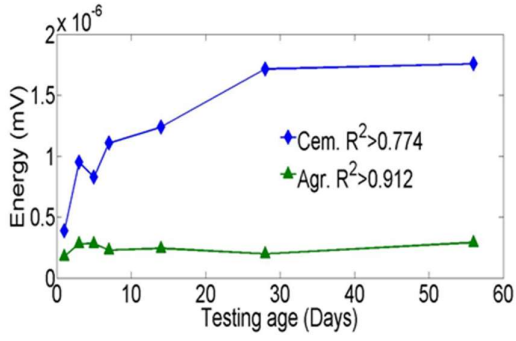
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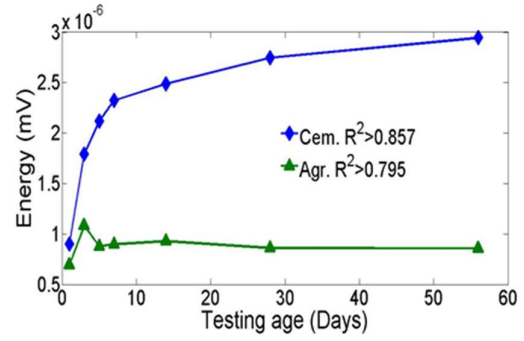
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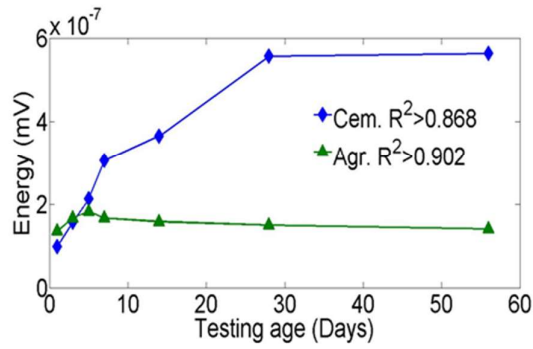
Fig. 6.



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Fig. 7.

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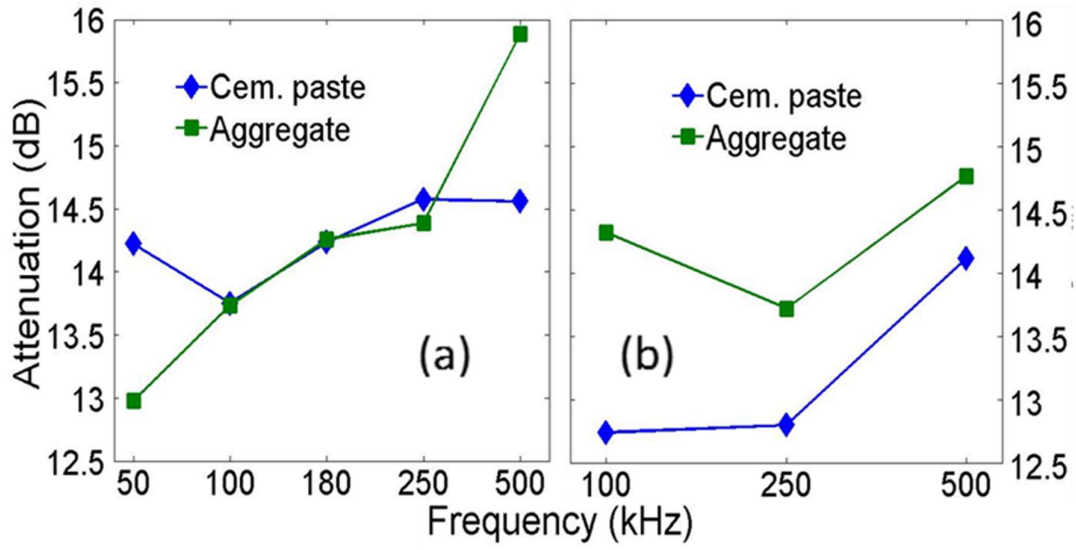


Fig. 8.

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593 Table 1. Chemical composition of main oxides for the Ordinary Portland Cement (OPC)  
594 used.

<b>Main oxides components</b>	<b>%</b>
<b>SiO<sub>2</sub></b>	21.1
<b>Al<sub>2</sub>O<sub>3</sub></b>	3.7
<b>Fe<sub>2</sub>O<sub>3</sub></b>	4.5
<b>CaO</b>	61.9
<b>MgO</b>	1.8
<b>K<sub>2</sub>O</b>	0.3
<b>Na<sub>2</sub>O</b>	0.1
<b>SO<sub>3</sub></b>	1

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609 Table 2. Wavelengths generated by the time-evolution of specimens, using the ultrasonic  
 610 pulse velocity of each transducer

<b>Transducer frequency (kHz)</b>	<b>Transducer diameter (D) (mm)</b>	<b>wavelength at day 1 (<math>\lambda</math>) (mm)</b>	<b>wavelength at day 56 (<math>\lambda</math>) (mm)</b>	<b>Mean <math>\pm</math> std dev wavelength (<math>\lambda_M</math>) (mm)</b>	<b>Radiation pattern <math>\beta = \frac{D}{\lambda_M}</math></b>
<b>Transversal wave transducer</b>					
<b>50</b>	42	53.2	60.0	56.9 $\pm$ 2.29	0.7
<b>100</b>	29	21.6	25.7	24.2 $\pm$ 1.56	1.2
<b>180</b>	42	13.7	13.5	13.6 $\pm$ 0.13	3.1
<b>250</b>	29	6.5	7.8	7.3 $\pm$ 0.42	4.0
<b>500</b>	29	3.2	3.8	3.6 $\pm$ 0.21	8.1
<b>Longitudinal wave transducer</b>					
<b>100</b>	42	22.5	28.5	25.5 $\pm$ 2.19	1.6
<b>250</b>	42	15.5	19.1	17.4 $\pm$ 1.51	2.4
<b>500</b>	29	7.3	10.1	9 $\pm$ 0.95	3.2

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620 Appendix A

621 Fig. 9. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
622 transducers used were 50 kHz transversal wave.

623 Fig. 10. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
624 transducers used were 180 kHz transversal wave.

625 Fig. 11. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
626 transducers used were 500 kHz longitudinal wave.

627 Fig. 12. Frequencies versus testing age for (a) aggregate and (b) cement paste. The  
628 transducers used were 500 kHz transversal wave.

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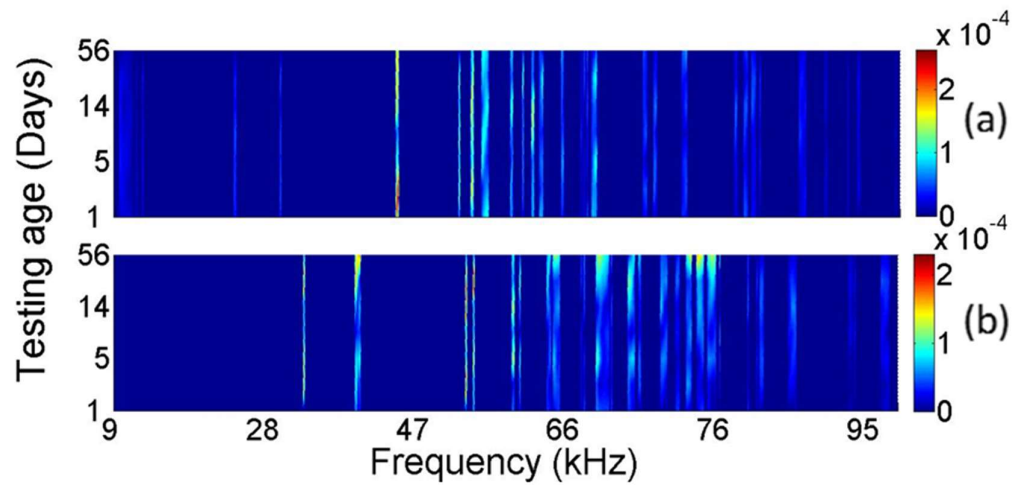


Fig. 9.

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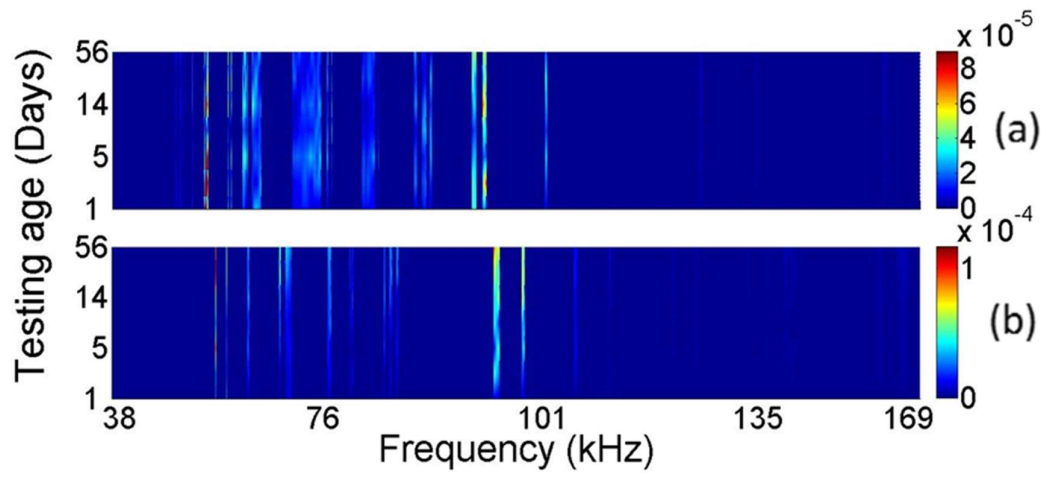


Fig. 10.

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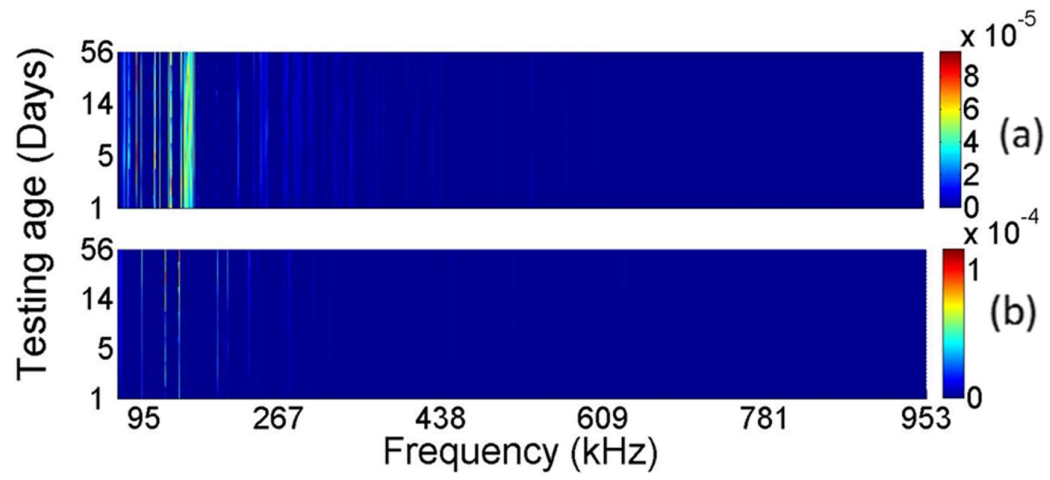


Fig. 11.

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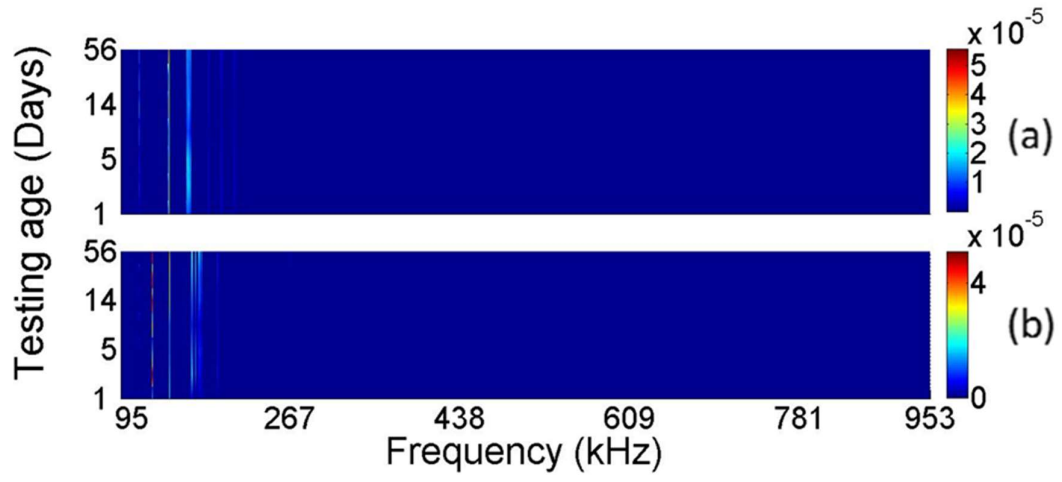
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Fig. 12.

1 ***Capítulo II***

2 Non-destructive ultrasound based cement paste carbonation index

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17 **Abstract**

18 This study presents a new method to assess the degree of carbonation in cement paste specimens,  
19 by analyzing their ultrasound response to different excitation frequencies. Eighteen cylindrical  
20 cement paste specimens measuring 50 mm in diameter and 57 mm in height were cast with three  
21 different water to cement ratios ( $w/c = 0.60, 0.50$  and  $0.40$ ). The samples were moist cured for  
22 four months to reach a high degree of hydration. Then the samples were conditioned in an  
23 environmental chamber at  $30\text{ }^{\circ}\text{C}$  and  $65\%$  relative humidity. After obtaining constant mass, each  
24 specimen was coated with epoxy resin on the curved surface, one circular face was covered with  
25 parafilm. The specimens were then exposed to accelerated carbonation in an environmental  
26 chamber at  $30\text{ }^{\circ}\text{C}$ ,  $65\%$  relative humidity, and  $4\%$   $\text{CO}_2$  (v/v). One control specimen was kept at  
27 the same environmental conditions of temperature and relative humidity, without  $\text{CO}_2$ . All the  
28 specimens were tested at 1, 3, 5, 7, 14, 28 and 56 days, during carbonation process, using  
29 longitudinal and transversal ultrasonic waves with frequencies ranging from 100 kHz to 500  
30 kHz. Assuming the carbonation in cement paste causes changes in the frequencies spectrum as  
31 the carbonation front progresses, causing important changes on porosity and pore structure of the  
32 cement paste. By analyzing these changes, we may infer the degree of carbonation in the  
33 specimen. Results show that it was possible to assess the degree of carbonation of cement paste  
34 specimens by frequency domain analysis as a function of carbonation time, regardless of the  
35 water to cement ratio.

36 **Keywords:** ultrasound analysis; carbonation; cement paste; Fourier analysis; carbonation  
37 index

## 38 **1. Introduction**

39 The alkaline pore solution of the hydrated cement paste protects the reinforcing steel from  
40 corrosion caused by carbonation [1,2]. Carbonation is the result of the reaction between CO<sub>2</sub>  
41 from the environment and the main hydration products such as calcium hydroxide and CSH in  
42 the presence of moisture, generating calcium carbonate and silica gel [3]. As a result, the pH of  
43 the pore solution is reduced from above 12.5 to less than 9. This may destroy the passivating  
44 layer on the steel and corrosion starts [4]. Carbonation not only affects the chemistry of the pore  
45 solution, but also affects the porosity and pore size distribution of the cement paste [5]. In some  
46 cases, the porosity is reduced and the strength is increased. In other cases, the porosity is  
47 increased, especially when using fly ash [6].

48 Penetration of CO<sub>2</sub> is function of the permeability of hardened cement paste in concrete, as well  
49 as the CO<sub>2</sub> concentration and environmental factors such as temperature and relative humidity  
50 [7]. The carbonation front follows the square root of time behavior. Regarding the environmental  
51 factors, the penetration is strongly influenced by the concentration of CO<sub>2</sub>, temperature and  
52 relative humidity [8]. Therefore, conclusions based on accelerated carbonation tests, can be used  
53 to estimate the service life of reinforced concrete structures [6, 9].

54 The most common and simple technique used to measure the carbonation depth is by spraying a  
55 phenolphthalein indicator on a freshly broken surface. However, it is destructive in nature and in  
56 addition it underestimates the carbonation front penetration [10]. Other destructive methods such  
57 as the thermogravimetric analysis (TGA), X-Ray diffraction and Fourier transformation infrared  
58 spectroscopy (FTIR) have been also used to detect the changes in concrete produced by

59 carbonation [11,12,13]. The gammadensimetry [14] is able to measure the total penetration of the  
60 CO<sub>2</sub> during laboratory accelerated tests. However, in order to obtain a full picture of the  
61 carbonation process numerous samples need to be tested.

62 Transmission wave velocity and wave attenuation are affected by the density and the elastic  
63 modulus of materials [15,16], which makes ultrasound wave analysis a potential technique for  
64 assessing carbonation of the cement paste. This article presents a new approach on the analysis  
65 of ultrasound signals to study carbonation of mature cement paste specimens with different w/c  
66 ratios. This approach was adapted from our methodology proposed for distinction of aggregate  
67 and cement paste in concrete samples from analysis of ultrasound signals [17].

## 68 **2. Experimental procedure**

### 69 **2.1. Materials**

70 Mexican ordinary Portland cement class 30RS (ASTM C150 Type I) was used. A total of 18  
71 cylindrical cement paste specimens measuring 50 mm in diameter and 57 mm in height were cast  
72 with three different w/c ratios of 0.60, 0.50 and 0.40. After one day, the specimens were stripped  
73 and moist cured in a saturated lime solution at  $40 \pm 2$  °C for four months, according to the  
74 ASTM C31 [18]. Then the specimens were stabilized in an environmental chamber at 30 °C and  
75 65 % relative humidity, until constant mass was achieved.

76 After stabilization, each specimen was coated with epoxy resin on the curved surface. Before  
77 carbonation commenced, one circular face was covered with paraffin film to expose only one  
78 face to CO<sub>2</sub> penetration. Five specimens of each w/c ratio were exposed to accelerated

79 carbonation in an environmental chamber at 30 °C, 65 % relative humidity, and 4 % CO<sub>2</sub> (v/v).  
80 One control specimen was kept at the same temperature and relative humidity, without CO<sub>2</sub>.

## 81 **2.2. Method**

### 82 **2.2.1. Ultrasonic measurements**

83 Six specimens were tested at ages of 1, 3, 5, 7, 14, 28 and 56 days, after beginning the  
84 carbonation process. For each test, the ends of each specimen were covered with paraffin film  
85 and heated with a heat gun for 60 seconds in order to obtain a smooth thin layer, which prevented  
86 penetration of the coupling agent (petroleum jelly) into the specimen. The excitation signal was  
87 generated by a 5077PR pulser/receiver with a voltage of 200 V and a variable factor of electronic  
88 damping or gain depending on the amplitude of the received signal. A record was maintained of  
89 their responses in through-transmission with longitudinal and transversal ultrasonic waves,  
90 generated by transducers with frequencies ranging from 100 kHz to 500 kHz. The position of the  
91 transducers and their applied pressure were kept constant during the experiment.

### 92 **2.2.2. FTIR measurements**

93 Three specimens of each w/c ratio were taken at 1, 3, 5, 7, 14, 28 and 56 days, for carbonated  
94 and control specimens. Once the epoxy resin coat was removed, the cement paste samples were  
95 crushed with hammer and chisel and then the small pieces were ground with an agate mortar.  
96 The fine powder obtained was sieved by the No. 100 mesh (150 µm), then oven-dried at 105 ° C  
97 for 24 hours, and finally stored in sealed bags. The bags were inside a desiccator with silica gel  
98 and soda lime at room temperature, to avoid moisture adsorption and carbonation of the powder  
99 samples until FTIR measurements were performed [13].

100 Three different FTIR spectra of each powder sample were taken with a ThermoScientific  
101 instrument Model Nicolet 6700 (ThermoScientific, MA, USA) and the mean value of the area  
102 under the curve was calculated for the vibration bands at  $1418\text{ cm}^{-1}$  and  $872\text{ cm}^{-1}$  corresponding  
103 asymmetric stretching and out plane bending for the carbonate ion in the  $\text{CaCO}_3$ . The initial  
104 amount of  $\text{CaCO}_3$  in the anhydrous cement (3%) was determined by thermogravimetric analysis  
105 and this value was related to the area under the curve of the bands corresponding to  $\text{CaCO}_3$  in the  
106 FTIR spectrum. This served as a basis for estimation of the amount of  $\text{CaCO}_3$  by weight in the  
107 carbonated samples.

### 108 **3. Wave analysis**

#### 109 **3.1. Signal processing**

110 The method proposed for analysis of the ultrasound signals was intended to assess the degree of  
111 carbonation in cement pastes with different w/c ratios. The main feature exploited by the  
112 proposed method was that the  $\text{CO}_2$  penetration affects the microstructure of the cement paste by  
113 generating of calcium carbonate, and the consequent modification of its density and physical  
114 properties [19]. Assuming that the degree of hydration [20] of the cement pastes was high  
115 because of the relatively long moist curing time, it was feasible to consider that variations in the  
116 ultrasonic response of cement pastes will be a result of carbonation, as opposed to the invariant  
117 behavior of the control specimens.

118 This analysis methodology was based on our previous work [17]. The first step in the proposed  
119 analysis was to calculate the time-discrete Fourier Transform (TDFT) of the ultrasound signal

120  $f(n\Delta t)$ , according to Eq. (1). The Fast Fourier Transform (FFT) is a standard signal processing  
121 technique for analyzing ultrasound signals [21].

$$122 \quad F(k\Delta\omega) = \frac{1}{2\pi} \sum_n f(n\Delta t) e^{-i(2\pi kn/N)\Delta t} \quad (1)$$

123 where T is the duration of the ultrasound signal, discretized at a sampling rate  $\Delta t=T/N$ , and  $k\Delta\omega$   
124 are the resulting discrete frequencies.

125 A window of  $F(k\Delta\omega)$  was subsequently selected from the frequency response function of the  
126 transducers with an intensity greater or equal to 6 dB, as specified by the ultrasound transducer  
127 manufacturer.

$$128 \quad G(k\Delta\omega) = \begin{cases} F(k\Delta\omega), & k_1 \leq k \leq k_2 \\ 0, & elsewhere \end{cases} \quad (2)$$

129 The changes over time of the seven spectra obtained (1, 3, 5, 7, 14, 28 and 56 days) from  
130 ultrasound measurements  $G_i(k\Delta\omega)$ , where  $1 \leq i \leq 7$ , was analyzed for the identification of the  
131 carbonation process. This analysis consists of the correlation of variations along time of every  
132 frequency amplitude  $|G_i(k\Delta\omega)|$ , with a constant function. For the identification of frequencies  $k_c$   
133 related to the carbonated cement paste. The correlation was made from the first two testing days,  
134 until the eighth day, obtaining seven sets of correlations for each specimen.

135 A histogram was obtained for each of the seven sets of correlations. The abscissa  $x$  was  
136 discretized by class and associated to each correlation coefficient, for each of the seven sets of  
137 correlations. The ordinate represents the number of frequencies in each class. The size of the  
138 class was chosen with a precision of 0.001. In order to improve understanding of the data

139 behavior, the histogram obtained was duplicated using a function which satisfies the next  
140 condition:

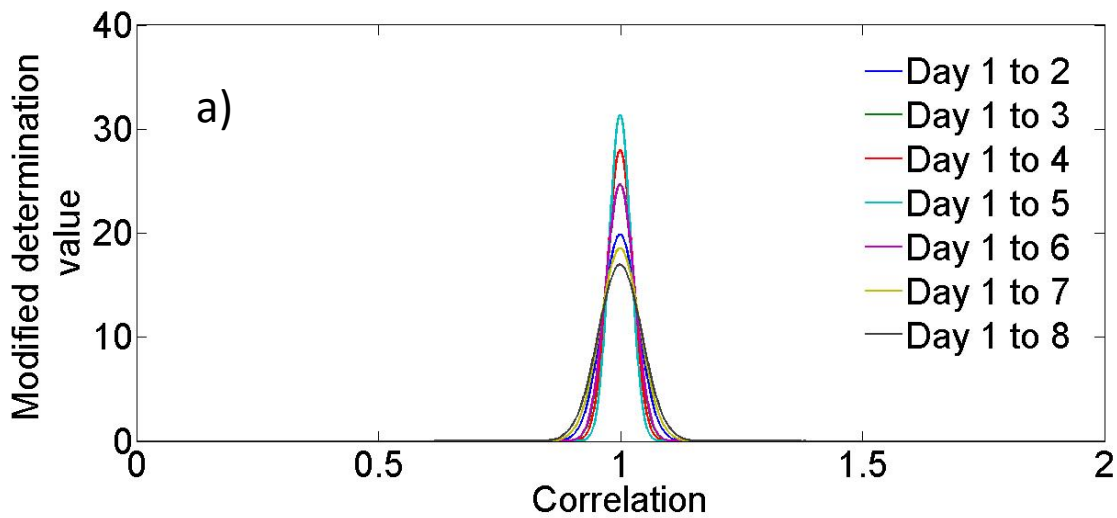
$$141 \quad H(x) = H(2 - x) \quad (3)$$

142 Such that the function obtained is defined as follows:

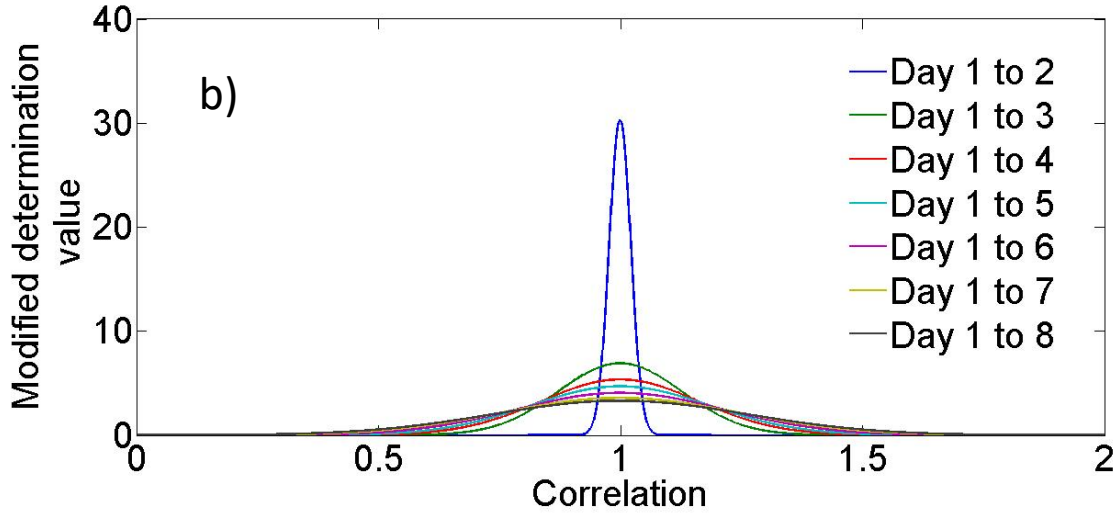
$$143 \quad K(x) = \begin{cases} H(x), & 0 \leq x \leq 1 \\ H(2 - x), & 1 < x \leq 2 \end{cases} \quad (4)$$

144 Then, the resulting signal was adjusted to a normal distribution function, minimizing the squared  
145 error, as shown in Figure 1.

$$146 \quad I_a(x) = \beta e^{-\left(\frac{x-1}{\sigma}\right)^2} \quad (5)$$



147



148

149 Figure 1 Normal distribution adjustment of the accumulative data for each measurement day, of  
 150 control (a) and carbonated (b) specimen, for the water to cement ratio of 0.50 using the  
 151 longitudinal wave transducer of 100 kHz.

152 This process was performed for each of the seven set of correlations. Using the inverse of the  
 153 amplitude  $\beta$ , the index  $I$  was obtained as follows.

$$154 \quad I = \frac{(w/c) \cdot \gamma}{\beta} \quad (6)$$

155 where  $I$  is the carbonation index, dependent on the  $w/c$  and the empirical obtained factor  $\gamma$ , that  
 156 may be linked to exposure conditions.

157 In previous studies, it was found that the exposure to high concentration of  $CO_2$  cause a  
 158 degradation of the calcium silicate hydrate (C-S-H), weakening the cement matrix which will  
 159 generate cracks through the specimens, being more evident with higher water/cement ratios [5].



160 Since the generation of cracks is highly probable when the cement paste is carbonated, as the  
161 products generated by the carbonation reaction occupy a higher volume than the hydration  
162 products [22, 23]. For this reason, it was necessary to correct the previous index by attenuation  $A$   
163 using the following equation.

$$164 \quad A = -\log\left(\frac{A_E}{A_S}\right) \quad (7)$$

165 where  $A_E$  is the amplitude of the first measurement day and  $A_S$  is the amplitude of the  
166 subsequent measurement days.

167 The correction by attenuation Eq. (7) of the index obtained in Eq. (6), produces the index  $I_A$  for  
168 the carbonation of the cement paste was obtained as:

$$169 \quad I_A = I - A \quad (8)$$

### 170 **3.2. Wavelength**

171 The wavelength  $\lambda$  depends on wave velocity (UPV) and transducer frequency ( $\Omega$ ) according to  
172 Eq. (9).

$$173 \quad \lambda = \frac{UPV}{\Omega} \quad (9)$$

174 The sensitivity to detect defects/changes in the cement paste was given by the relation between  
175 the size of the defect/changes and the wavelength [24]. The sensitivity depends on the transducer  
176 frequency, which increase directly proportional to the frequency employed, assuming constant

177 UPV [25]. In carbonated specimens, different degrees of carbonation may be detected depending  
178 on the frequency transducers used.

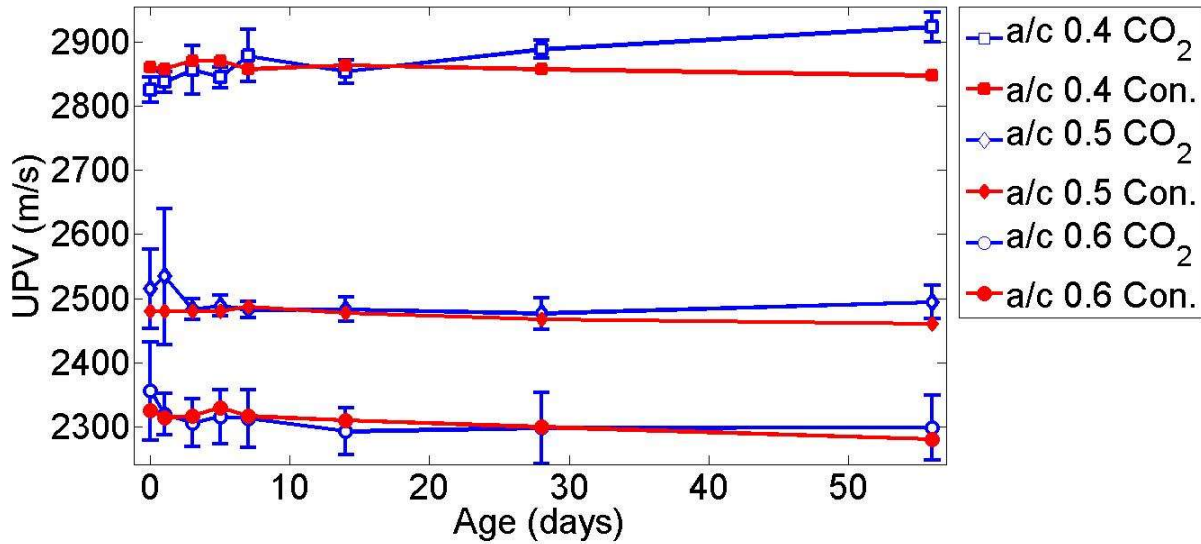
## 179 **4. Results and discussion**

### 180 **4.1. Ultrasound pulse velocity**

181 The first analysis of the data was calculating the ultrasound pulse velocity (UPV). In Figure 2 it  
182 is shown the evolution in time of the UPV for all the specimens with three different w/c ratios.

183 The UPV allows only distinction between the different w/c ratio employed, but not for the  
184 difference between control and carbonated specimens.

185 As can be seen, the standard error of the UPV for the w/c = 0.50 and 0.60 was higher in  
186 comparison with that of the w/c = 0.40. This effect may be due to the generation of cracks caused  
187 by the carbonation process which was higher for the w/c = 0.50 and 0.60. In the case of the w/c =  
188 0.40 showed an increase of the UPV after 30 days of CO<sub>2</sub> exposure, influenced by the attack of  
189 C-S-H and CH producing important changes in the elastic modulus and density, despite of the  
190 short carbonation front.



191

192 Figure 2. Evolution in time of the UPV versus testing age for control (red lines) and carbonated  
 193 (blue lines) specimens for water/cement ratios of 0.40 (square), 0.50 (diamonds) and 0.60  
 194 (circles). The error bars represent one standard deviation.

195 Results from ultrasound testing were compared with the content of calcium carbonate (CaCO<sub>3</sub>)  
 196 estimated by the Fourier Transform Infrared Spectroscopy (FTIR) measurements [13], presented  
 197 in the following section.

198 **4.2. Carbonation index**

199 Variations in the porosity and pore structure of cement pastes caused by carbonation depend on  
 200 the properties of the cement paste and the exposure environmental conditions [19, 27]. Three w/c  
 201 ratios (0.40, 0.50 and 0.60) were analyzed for indexing the degree of carbonation in cement  
 202 pastes whose results are shown in Figures 3-5.

203 It was expected that the progressive changes caused by the carbonation process, gradually  
204 modify the cement paste microstructure. These variations slowly affect the frequency of the  
205 ultrasound response. This effect distinguishes the control from the carbonated specimens.

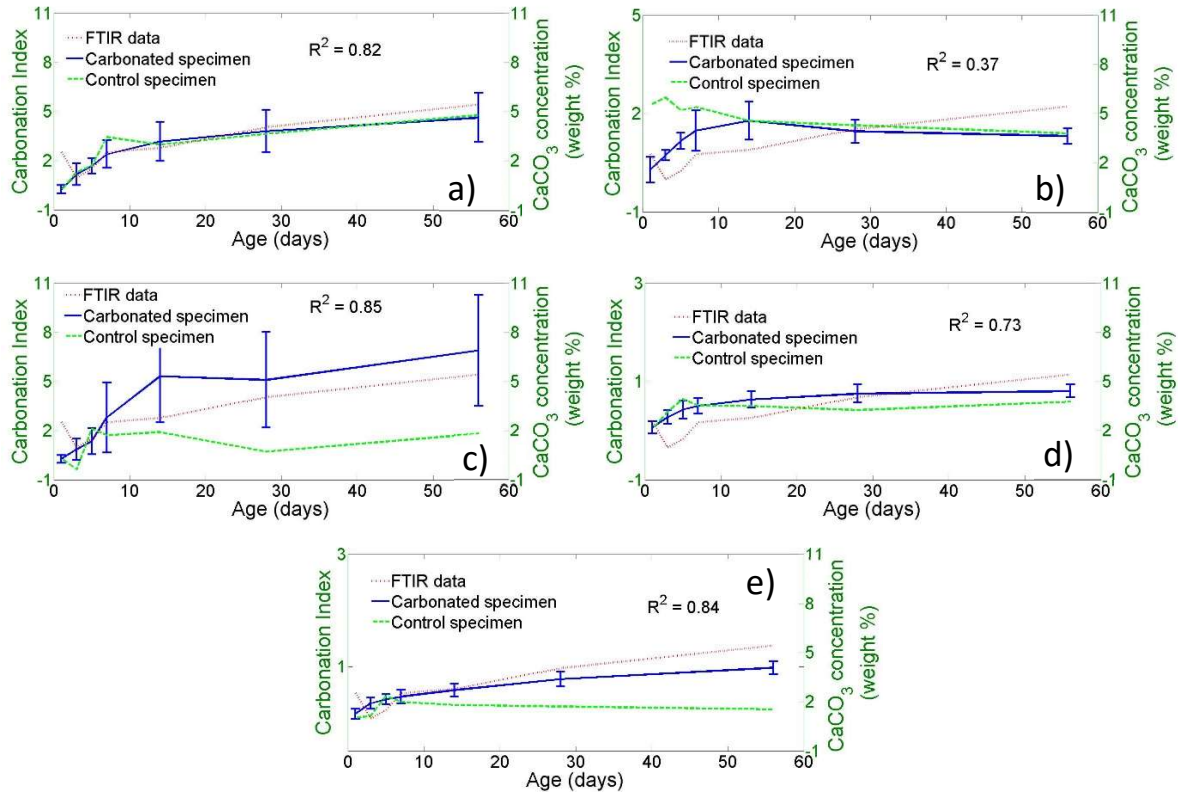
#### 206 **4.2.1. Cement paste samples with water/cement ratio of 0.60**

207 Figure 3 shows the results of the ultrasonic measurements for the control and carbonated  
208 specimens for cement paste with w/c ratio of 0.60. Figure 3a shows the evolution over time of  
209 the carbonation index for the control and carbonated specimen tested with 100 kHz longitudinal  
210 wave transducers. The behavior between both types of specimens was similar, although the trend  
211 of the carbonation index had a strong correlation with the FTIR data. The behavior of the control  
212 specimen may be explained by the stabilization process which cause drying shrinkage and  
213 cracking of the cement paste [28]. This was reflected in variations of the ultrasound response of  
214 the control specimens during the measurements as had been seen in previous studies [29,25].  
215 This effect was present in the rest of the measurements with other transducers, as can be seen in  
216 Figures 3b-e, but with less impact in the index response due to the variations in wavelength  
217 calculated in Table 1 [30].

218 In Figures 3b and 3d, measurements with the 100 kHz and 250 kHz transversal wave transducers,  
219 respectively, had a weak correlation with the FTIR data. The former correlation was mostly  
220 caused by the cracking and porosity of the specimens with a high w/c ratio [19]. These factors  
221 had a high impact on the propagation of transversal waves, making it difficult to assess the  
222 carbonation penetration, since they cannot propagate through fluids (water and air) [31].

223 Figure 3c shows the measurements with the 250 kHz longitudinal wave transducers. The control  
224 specimens keep a behavior almost constant in contrast with the index of the carbonated specimen  
225 whose values increase. The correlation between the carbonation index proposed and the FTIR  
226 data was strong. Although, the correlation was weaker than that obtained with the 100 kHz  
227 longitudinal wave transducers, in this case, it was possible to distinguish between the behavior of  
228 the control and the carbonated specimen. This distinction was possible due to the changes in  
229 wavelength and in the propagation properties of this kind of waves (Table 1) which can travel  
230 through solids and fluids [32].

231 Measurements with the 500 kHz longitudinal wave transducers are shown in Figure 3e. The  
232 distinction between control and carbonated specimen behavior was easier, in comparison with  
233 the results of the other transducers. Also, the carbonation index had the strongest correlation with  
234 the FTIR data and the dispersion of the carbonation specimens was the lowest. The wavelength  
235 and the propagation properties of the longitudinal waves of these transducers [33], as shown in  
236 Table 1, improves tracking of the carbonation process. Furthermore, the final value of the  
237 carbonation index proposed was close to the one obtained with FTIR. However, the losses of  
238 energy are so high that causes underestimation of carbonation degree.



239

240 Figure 3. Evolution of carbonation index and content of calcium carbonate by FTIR, versus  
 241 testing age for longitudinal wave transducers of 100 kHz (a), 250 kHz (c), 500 kHz (e) and  
 242 transversal wave transducers of 100 kHz (b), 250 kHz (d), for cement paste specimens with w/c  
 243 ratio = 0.60.

244 Table 1 shows how the wavelength of the carbonated specimens was lower than the control  
 245 samples, which is explained by the density increase due to generation of calcium carbonate  
 246 ( $\text{CaCO}_3$ ), for the same frequency of the transducers. As the transducer frequency increases, the  
 247 wavelength decreases enhancing the sensitivity of the ultrasound signals, along with reducing the  
 248 standard deviation between specimens. Also, a small decrease between the same frequency  
 249 transducers but different wave type is produced; this decrease was more visible in carbonated  
 250 specimens.

251 Table 1. Mean and standard deviation of the wavelengths generated by the time-evolution of  
 252 control and carbonated specimens with w/c=0.6

Frequency (kHz)	Diameter of Transducer (D) (mm)	Control wavelength ( $\lambda$ ) (mm)		Carbonated wavelength ( $\lambda$ ) (mm)	
		Mean ( $M_{\lambda_{Co}}$ )	Std dev	Mean ( $M_{\lambda_{Ca}}$ )	Std dev
Longitudinal wave transducer					
100	42	23.13	0.47	23.12	0.15
250	42	9.38	0.31	9.28	0.08
500	29	4.74	0.17	4.68	0.04
Transversal wave transducer					
100	29	22.56	0.66	22.33	0.17
250	29	9.35	0.34	9.21	0.09

253

#### 254 4.2.2. Cement paste samples with water/cement ratio of 0.50

255 Figure 4 shows the results of the ultrasonic measurements for the control and carbonated  
 256 specimens with a w/c ratio of 0.50. The results of the ultrasound measurements with the 100 kHz  
 257 transversal wave transducer are shown in Figure 4b. Although, the correlation with the FTIR data  
 258 was weak, the distinction between the control and the carbonated specimens was possible. This  
 259 was because of the slow penetration of CO<sub>2</sub>, result of the lower permeability of this paste  
 260 compared with the permeability of the paste with w/c = 0.60 [34].

261 This decrease in the carbonation ratio allows a slower formation of the CaCO<sub>3</sub>, enlarging the  
 262 time of occurrence of the structural changes in the cement paste [19]. These smooth the process  
 263 of carbonation allowing a better assess of the changes in the cement matrix and therefore a better  
 264 studied of the specimen, which was reflected in the stronger correlation of the carbonation index  
 265 for all transducers, in comparison with the cement paste of 0.60. In the case of Figures 4a, 4c, 4d,  
 266 and 4e, the correlation with the FTIR data becomes stronger as the transducer frequency

267 increases. This behavior was related to the gain in sensitivity due to the decrease of the  
268 wavelength as the transducer frequency increases, as shown in Table 2.

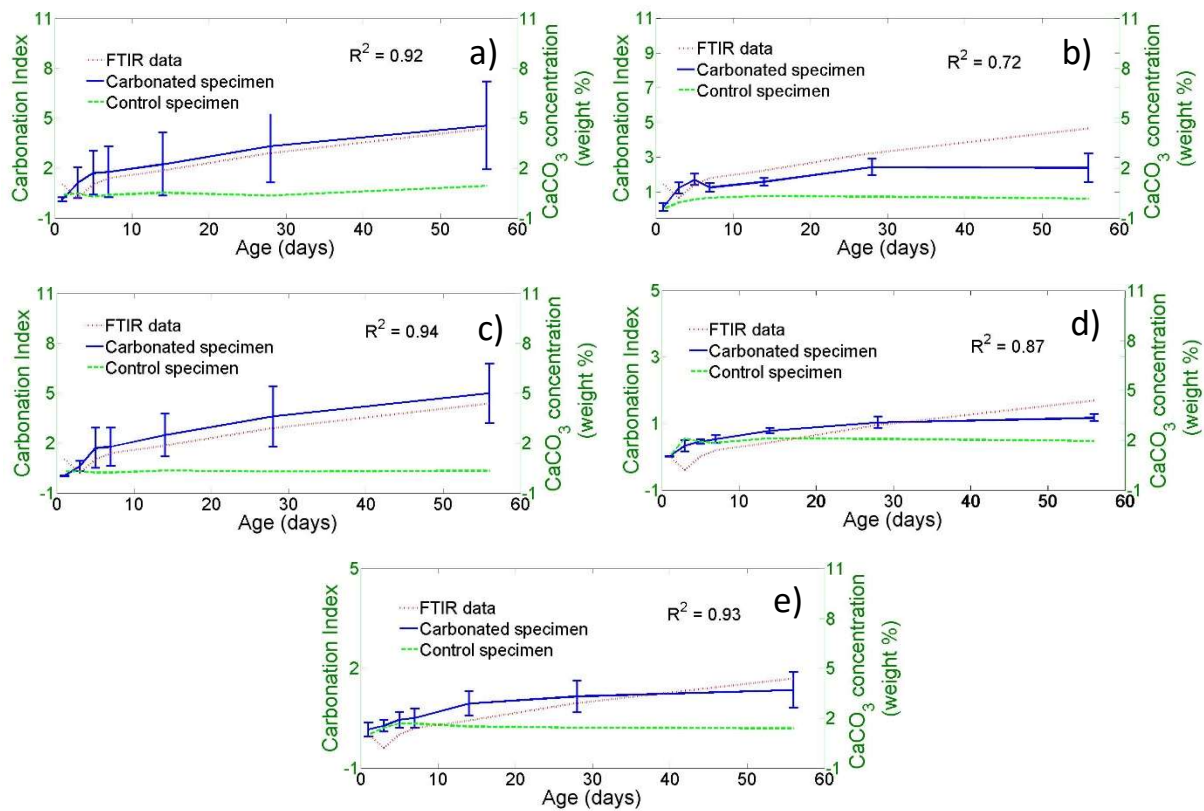
269 Figure 4a shows the evolution in time of the carbonation index for the 100 kHz longitudinal  
270 wave transducers. It was possible to distinguish the behavior between the control and the  
271 carbonated cement paste specimens. The correlation was stronger than the 100 kHz transversal  
272 wave transducer, which might be explained by fewer losses of the signal energy by cracks, void  
273 and water caused by the carbonation process [6]. Nevertheless, the standard error of the data was  
274 higher than the ones of Figures 4c, 4d and 4e.

275 The correlation obtained by the 250 kHz longitudinal wave transducer was the strongest on  
276 evaluating the 0.5 w/c cement paste, as shown in Figure 4c. The rise of the correlation was  
277 mostly caused by the decrease of the wavelength and the property of the longitudinal waves to  
278 travel through different media, as shown in Table 2. This causes less energy losses due to the  
279 specimen heterogeneities with better interaction during the measurements, but increasing the  
280 standard error between the carbonation index for different specimens, as well as the sensitivity.

281 The results for the 250 kHz transversal wave transducer are shown in Figure 4d. The stronger  
282 correlation, compared with the 100 kHz transversal wave transducer, was due to the lower  
283 wavelength. However, the sensitivity was not enough to surpass the correlation of higher or  
284 equal longitudinal wave transducers frequency. Although, the preference of the transversal waves  
285 to propagate through solids and the slower generation of the solid phases ( $\text{CaCO}_2$ ) of the  
286 carbonation process had high impact on the standard error of the carbonation index of different  
287 specimen measurements, as seen in Figure 4d.



288 The results for the 500 kHz longitudinal wave transducer are shown in Figure 4e. The correlation  
 289 with the FTIR data was the second strongest of the transducer measurements, because of the  
 290 wavelength (Table 2) which allows a better interaction with the specimen recollecting more  
 291 information of the specimen, also cause a higher loss of energy. As expected, the sensitivity  
 292 increases with time, but the heterogeneities between different specimens induces a rise of  
 293 standard error in the carbonation index for different specimens, caused by the heterogeneities  
 294 produced by the carbonation process.



295  
 296 Figure 4. Evolution of carbonation index and content of calcium carbonate by FTIR, versus  
 297 testing age for longitudinal wave transducer 100 kHz (a), 250 kHz (c), 500 kHz (e) and  
 298 transversal transducer 100 kHz (b), 250 kHz (d) for cement paste specimens with w/c ratio =  
 299 0.50.

300 As stated, the wavelength decreases as the frequency transducer increases, but the wavelength  
 301 between control and carbonated specimens was very close one to another, regardless of the  
 302 transducer frequency. This behavior was due to the slower carbonation reaction process. The  
 303 decrease between the same frequency but different type of wave was more evident with the 100  
 304 kHz transducers than with the 250 kHz transducers.

305 Table 2. Wavelengths generated by the time-evolution of control and carbonated specimens,  
 306 using the ultrasonic pulse velocity of each transducer for a water/cement ratio of 0.50.

Frequency (kHz)	Diameter of Transducer (D) (mm)	Control wavelength ( $\lambda$ ) (mm)		Carbonated wavelength ( $\lambda$ ) (mm)	
		Mean ( $M_{\lambda Co}$ )	Std dev	Mean ( $M_{\lambda Ca}$ )	Std dev
Longitudinal wave transducer					
100	42	28.63	0.23	28.60	0.08
250	42	10.08	0.13	9.96	0.06
500	29	5.07	0.03	5.01	0.02
Transversal wave transducer					
100	29	24.16	0.24	24.16	0.13
250	29	10.00	0.16	9.88	0.06

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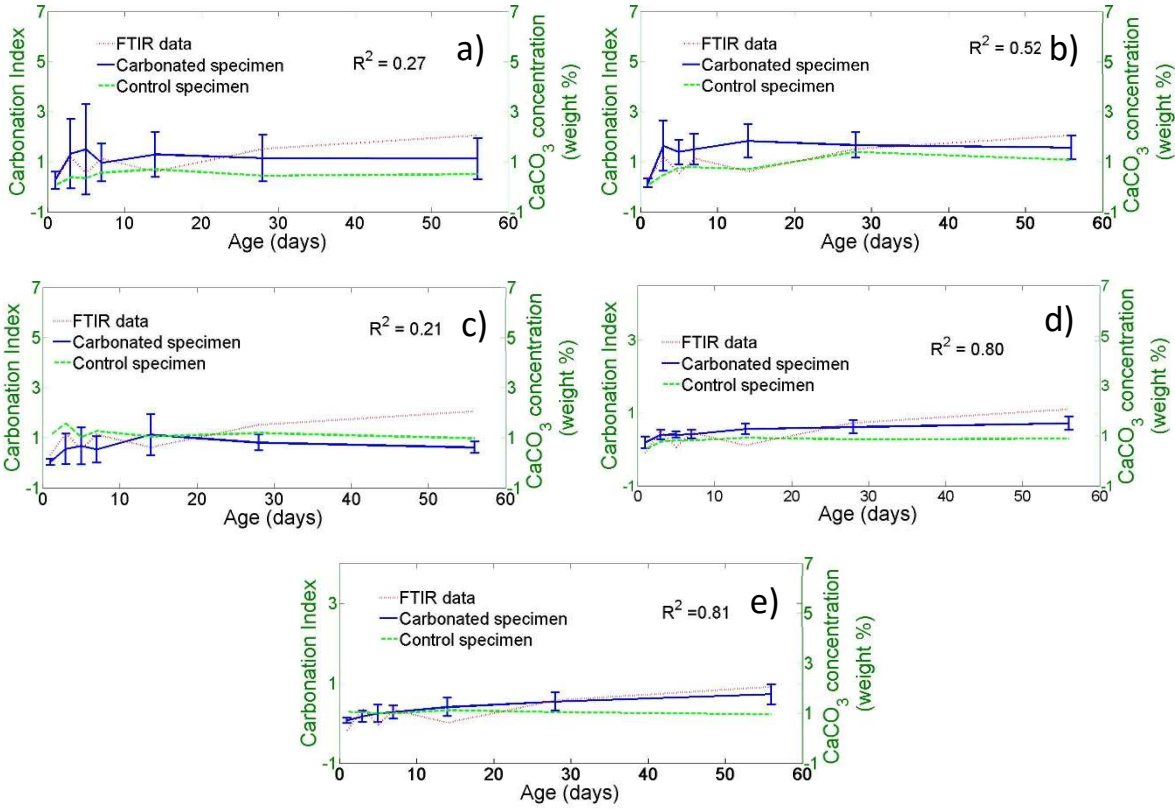
#### 308 4.2.3. Cement paste samples with water / cement ratio of 0.40

309 Figure 5 shows the results of the ultrasonic measurements for the control and carbonated  
 310 specimens with w/c ratio = 0.4. As can be seen in Figure 5a- 5c, it was difficult to distinguish  
 311 between control and carbonated specimens for this w/c ratio, neither using longitudinal wave  
 312 transducers of frequencies of 100 and 250 kHz, nor 100 kHz transversal wave transducers. The  
 313 behavior observed of the carbonation index was rather constant in comparison with the results of  
 314 the previous w/c ratio analyzed. Even with the decrease of the wavelength, it cannot be assessed  
 315 the carbonation degree of the specimens. The lower penetration in these specimens was caused

316 by the porosity reduction and the carbonation reaction on the surface of the specimens which  
317 prevent the penetration of CO<sub>2</sub> [19].

318 The correlation gets stronger between the FTIR data and the energy percentage calculated for the  
319 250 kHz transversal wave transducer. The formation of solid calcium carbonated in the cement  
320 paste produced a higher interaction with the propagation of transversal waves, since they prefer  
321 to propagate through solid media [30]. This preference allows the wave for better detection of  
322 changes caused by the carbonation process in comparison with the longitudinal wave transducer  
323 of the same frequency (even with similar wavelengths, Table 3), enabling the distinction between  
324 the control and the carbonated cement paste even if the carbonation was at onset.

325 In Figure 5e, the results of the 500 kHz longitudinal wave transducer are shown. It was possible  
326 to distinguish the behavior between the control and the carbonated specimen, since the  
327 wavelength was the smallest produced (Table 3). The correlation with the FTIR data was  
328 stronger than with the others frequencies [13]. As stated, this was due to the size of the  
329 wavelength and also to the interaction of the longitudinal wave with the varying media.  
330 However, the standard error of the carbonation index increases with time because of the high  
331 sensitivity of the high frequency transducer.



332

333 Figure 5. Evolution of carbonation index and content of calcium carbonate by FTIR, versus  
 334 testing age for longitudinal wave transducer 100 kHz (a), 250 kHz (c), 500 kHz (e) and  
 335 transversal transducer 100 kHz (b), 250 kHz (d) for cement paste specimens with w/c ratio =  
 336 0.40.

337 Variations in wavelength are shown in Table 3. Although, the wavelength of the control and  
 338 carbonated specimens was very similar, the small changes in the wavelength are caused by the  
 339 carbonation process. Also, the wavelength of the different wave type transducers and same  
 340 frequency are very close one to another.

341

342 Table 3. Wavelengths generated by the time-evolution of control and carbonates specimens,  
 343 using the ultrasonic pulse velocity of each transducer for a water/cement ratio of 0.40.

Frequency (kHz)	Diameter of Transducer (D) (mm)	Control wavelength ( $\lambda$ ) (mm)		Carbonated wavelength ( $\lambda$ ) (mm)	
		Mean ( $M_{\lambda Co}$ )	Std dev	Mean ( $M_{\lambda Ca}$ )	Std dev
Longitudinal wave transducer					
100	42	28.63	0.23	28.60	0.08
250	42	11.55	0.07	11.53	0.03
500	29	5.82	0.06	5.83	0.01
Transversal wave transducer					
100	29	27.67	0.20	27.97	0.28
250	29	11.43	0.10	11.50	0.04

344  
 345 As can be seen in Table 3, the wavelength between the w/c of 0.50 and 0.40 was almost the  
 346 same, but in the w/c of 0.60, the wavelength was smaller. This was an indicator of the higher  
 347 porosity of these specimens [24]. The close value of the control wavelength to the carbonated  
 348 specimen wavelength, can explain the atypical results shown in Figure 3a, compared with the  
 349 results shown in Figure 4a, where a clear differentiation between carbonated and control  
 350 specimen can be observed. In the case of Figure 5a, the sensitivity was not enough to  
 351 differentiate between the control and the carbonated specimen. However, unlike the Figure 3a,  
 352 both behaviors remained constant, which was an indication that the carbonation front had not  
 353 advanced enough to be detected and the response behavior seems to be constant, contrary with  
 354 the results of Figure 3a, where the control specimen had an abnormal behavior.

355 In Figure 3b, the abnormal behavior of the control specimen was even more obvious since the  
 356 index associated to it had significantly atypical, showing values below zero, which are  
 357 inconsistent with the rest of the results. In Figure 4b the differentiation between control and  
 358 carbonated specimen was better than in the rest of the measurements. The increase of the

359 wavelength, observed with the 100 kHz transversal wave transducers, is a parameter that  
360 indicates an increasing density of the specimens, with the reduction of the w/c ratio, better than  
361 the longitudinal wave transducer of the same frequency, as showed in Table 1-3.

362 Figure 3c, show an improvement in the distinction between the control and carbonated specimen  
363 almost as good as Figure 4c, where a clearer distinction between the carbonated and control  
364 specimen was possible, compere with Figure 5c. As the transducer frequency employed rise the  
365 distinction between the cement past was more clear, as the gap between wavelength increase,  
366 being the biggest one among the w/c ratio of 0.6 and 0.4.

367 As can be seen in Figure 4d and 5d, the evaluation of the carbonation improves, even for the w/c  
368 ratio of 0.40, which before were unresponsive, except for the w/c ratio of 0.60. Both cases are  
369 related to the decrease of the wavelength, as can be seen in Table 1-3, also the asses of the degree  
370 of carbonation had a strong correlation to the one obtained with the FTIR results.

371 In the case of the results show in Figure 4e and 5e, the carbonation index had the strongest  
372 correlation with the FTIR data of all measurements. This behavior was due to the small  
373 wavelength, as show in table 1-3, and the property of the longitudinal waves to travel true all  
374 media, which allows the ultrasound wave, to interact with a larger amount of material without  
375 losing energy, enabling a better assess of the carbonation degree, regardless of the w/c ratio.

376

377

378

379 **5. Conclusions**

380 The following conclusions are drawn based on the results obtained in this paper:

- 381 1. The non-destructive carbonation index based on ultrasound measurements, allows to  
382 determine the carbonation degree of the cement pastes studied ( $w/c = 0.60, 0.50$  and  
383  $0.40$ ), assuming that the carbonation products will progressively change the frequency  
384 spectra in time. The observed trends in the carbonation index are in agreement with the  
385 FTIR intensities observed for  $\text{CaCO}_3$  generation.
- 386 2. For the cement pastes with  $w/c = 0.40$ , only the 250 kHz transversal wave transducer  
387 and the 500 kHz longitudinal wave transducer permit detection of the carbonation.
- 388 3. The carbonation index associated to  $w/c = 0.50$  had a strong correlation with the FTIR  
389 results, especially with the 100 and 250 kHz longitudinal wave transducers.
- 390 4. The 250 kHz longitudinal wave transducers were able to distinguish between the  
391 carbonated and control specimens, as well as to assess the degree of carbonation in the  
392  $w/c = 0.60$ , despite the atypical behavior of the control specimen.

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1 ***Capítulo III***

2 Ultrasound determination of chloride presence in cement paste

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17 **Abstract**

18 Durability of concrete structures depends on ability of the cover concrete to resist penetration of  
19 harmful substances. Chloride ingress disrupts the passive film formed on the steel surface  
20 causing corrosion. Traditional techniques used to determine chloride penetration involve  
21 destructive sampling and chemical analysis. In order to improve some properties of concrete,  
22 Portland cement is partially substituted with mineral additions such as silica fume and fly ash.  
23 This study is focused on the ability of longitudinal and transversal ultrasound waves to detect the  
24 presence of chlorides in cement paste specimens containing mineral admixtures. Eighteen  
25 cylindrical specimens measuring 65 mm in diameter and 100 mm in height were cast with a w/c  
26 ratio of 0.55, with 100% of ordinary Portland cement, 20% and 40% substitution of cement by  
27 class F fly ash and 10% substitution by silica fume. The specimens were moist-cured in a  
28 saturated lime solution at  $23 \pm 2$  °C for 120 days to reach a high degree of hydration. After oven-  
29 drying to constant mass, three samples of each group were vacuum saturated (4.08 bar) with  
30 NaCl solutions of 0%, 16.5 % (2.8 M), and 33 % (5.6 M). Ultrasonic measurements were  
31 undertaken at 0, 1, 3, 7 and 14 days using ultrasonic transducer with frequencies ranging from 50  
32 kHz to 2250 kHz. Results show that it is possible to distinguish between the samples with and  
33 without chlorides regardless of the mineral addition and between the mineral additions  
34 depending on its binding capacity.

35 **Keywords:** Chloride binding, ultrasound, fly ash, silica fume

## 36 **1. Introduction**

37 Chloride penetration leads to corrosion of the reinforcing steel in concrete structures [1].

38 Chlorides may already be present in the materials employed for the elaboration of concrete but  
39 normally they ingress from external sources such as marine environments or deicing salts [2].

40 Cement hydration products react to some extent with chlorides to form Friedel's salt, leaving  
41 some free chloride ions that are responsible for the steel corrosion [3,4]. Therefore, the chloride  
42 binding capacity of cement paste is an important factor that influences the durability of concrete  
43 structures exposed to chloride-rich environments [5,6].

44 The use of mineral admixtures, such as fly ash and silica fume, increases the chloride binding  
45 capability of cement paste depending on the degree of substitution of Portland cement [7].

46 Chloride ion ingress into the cover concrete occurs mainly by diffusion due to a concentration  
47 gradient. The rate of diffusion is controlled by the chemical and physical characteristics of the  
48 pore structure [8]. Chloride ingress also depends on the surface concentration, as well as  
49 environmental factors such as temperature and relative humidity [9], producing a non-uniform  
50 distribution of chloride ions in concrete [10]. In the case of partially saturated cover concrete,  
51 water will be drawn into the pore structure driven by capillary forces [11].

52 Determination of chloride ion content in concrete is performed by different methods such as  
53 Accelerated Chloride Penetration [12], which is used to determine the apparent coefficient of  
54 chloride ion diffusion into concrete. A reliable chloride concentration profile for high quality  
55 concrete is difficult to obtain, as the penetration depth may be insufficient at 90 days [13].

56 The Electrical Indication of Concrete Ability to Resist Chloride Ion Penetration [14] is a rapid  
57 test to determine concrete relative resistance to penetration chloride ions. The electrical  
58 resistivity assesses chloride ingress considering that chloride presence increases the electrical  
59 conductivity of concrete. Since electrical current is used, it affects all ions in the pore solution,  
60 not just chloride ions. Also, the measurements are made before a steady migration is achieved  
61 and the high voltage increases the temperature that affects the results.

62 Another method is the Electrical Migration Technique [15] which accelerates the movement of  
63 chloride ions by the use of a low intensity electrical field. Despite of reducing the temperature  
64 related to the use of an electrical current, there is still a significant drawback due to the  
65 employment of an electrical migration test.

66 The Ion Selective Electrode Method [16] determines free chlorides with an embedded sensor that  
67 assesses changes in potential related to chemical activity. Unfortunately, the range of evaluation  
68 is too narrow to assure a reliable measurement. These non-destructive methods are very sensitive  
69 to the moisture content and require a calibration relationship to ensure repeatability [17].

70 Ultrasound techniques such as ultrasound pulse velocity have been employed to detect the  
71 presence of internal defects in concrete structures. Low ultrasound frequency (50 kHz) are used  
72 for characterization of large defects [18]. Higher frequencies (0.5-1 MHz) have been  
73 demonstrated to be more suitable for detection of smaller variations related to concrete  
74 degradation [19]. The ultrasound pulse velocity is capable of detecting microstructural changes.  
75 The presence of cracks or aggregates in concrete strongly influence the scattering when the  
76 wavelength is of the same order, diminishing the sensitivity to such defects/components [20].



77 In this study, a signal processing technique of longitudinal and transversal waves produced at  
78 different frequencies was used for detecting the presence of chlorides in cement pastes with and  
79 without mineral additions. In order to increase the surface exposed to chloride binding,  
80 specimens were oven-dried to evaporate the free water and replace it with NaCl solutions at  
81 different concentrations.

## 82 **2. Experimental procedure**

### 83 **2.1. Materials**

84 Mexican ordinary Portland cement class 30RS (ASTM C150 Type I) was used. Eighteen  
85 cylindrical specimens measuring 65 mm in diameter and 100 mm in height were cast with a w/c  
86 ratio of 0.55, with 100% of ordinary Portland cement (OPC), 20% and 40% substitution of  
87 cement by class F fly ash (20FA and 40FA, respectively) and 10% substitution by silica fume  
88 (10SF). The cylinders were filled in two layers, compacted by applying sixty strokes on the base  
89 of the mold on each layer. Then the specimens were placed in a device designed to rotate the  
90 samples at a speed of 7 rpm to avoid sedimentation. After 1 day, the specimens were stripped and  
91 moist-cured in a saturated lime solution at  $23 \pm 2$  °C for 120 days to ensure complete hydration,  
92 according to the standard ASTM C31 [21].

93 In order to have similar initial conditions, all specimens were vacuum saturated with deionized  
94 water for at least one hour to ensure full saturation. Then the specimens were weighed and placed  
95 in an oven at 105 °C for two days, until constant mass is achieved. Once the specimens reached  
96 room temperature, three specimens of each group were vacuum saturated (4.08 bar) with NaCl

97 solutions of 16.5 % (2.8 M), 33 % (5.6 M) and one with only deionized water. All samples were  
98 covered with parafilm to prevent moisture loss.

## 99 **2.2. Method**

### 100 **2.2.1. Ultrasonic measurements**

101 To perform ultrasonic measurements, transducers were placed and centered on the circular faces.  
102 The pressure exerted on the transducers was controlled by a constant mass of 2.5 kg. The same  
103 position of the transducers was kept in through transmission measurements at different times.  
104 Ultrasonic measurements were made at 0, 1, 3, 7 and 14 days with transducer frequencies  
105 ranging from 50 kHz to 2250 kHz, for the three solutions used (NaCl 0% (deionized water),  
106 16.5% NaCl and 33% NaCl).

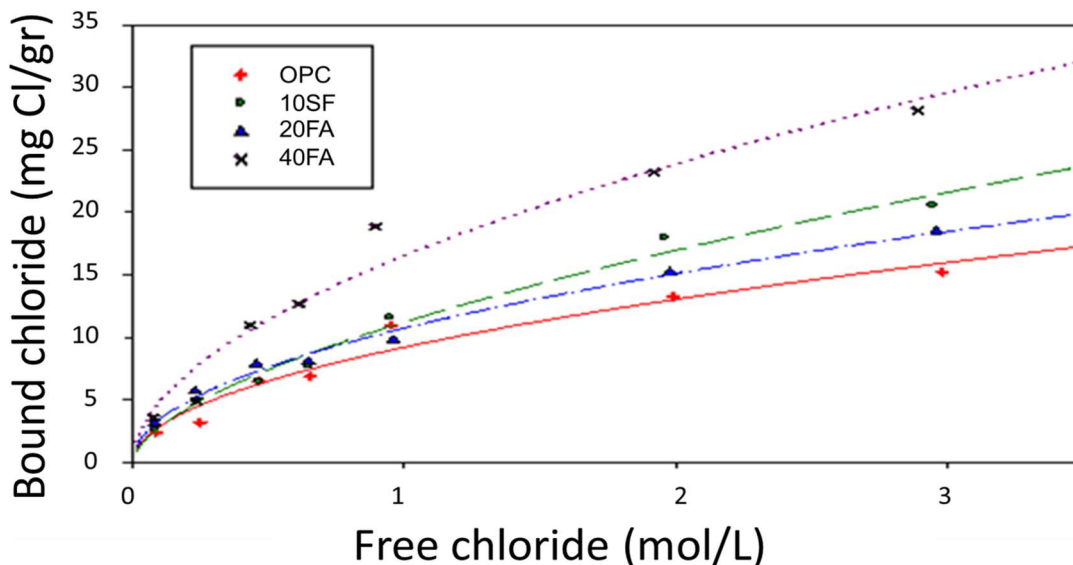
### 107 **2.2.2. Determination of chloride binding isotherms by the equilibrium method**

108 This method assumes that once the equilibrium between the external solution and the pore  
109 solution of the sample is reached, the reduction in the concentration of the external solution is  
110 due to chlorides bound by cement hydration products. Therefore, by knowing the initial and final  
111 concentrations, the volume of the external solution and the dry mass of the sample, it is possible  
112 the calculation of the amount of bound chlorides according to [22]:

$$113 \quad C_b = \frac{35.453 (C_i - C_e)}{w_{11}(1 - \xi_{11})} \quad (1)$$

114 where  $C_b$  is the amount of bound chlorides in mg  $Cl/$  per sample,  $V$  is the volume of the  
115 external solution in mL,  $C_i$  is the external solution initial concentration of chlorides in mol / L,  $C_e$

116 is the concentration of free chlorides in equilibrium with the external solution in mol / L, 35,453  
 117 is the molar mass of chloride ion,  $w_{11}$  is the mass in grams of the sample at a relative humidity  
 118 (RH) of 11% and  $\xi_{11}$  is the evaporable water content to 11% RH. The chloride isotherms for  
 119 each cementitious system are shown in Fig. 1.



120  
 121 Fig. 1. Chloride binding isotherms for the four cementitious systems at a temperature of 23 °C  
 122 for different chloride concentrations [23].

### 123 3. Wave analysis

#### 124 3.1. Ultrasound signal processing

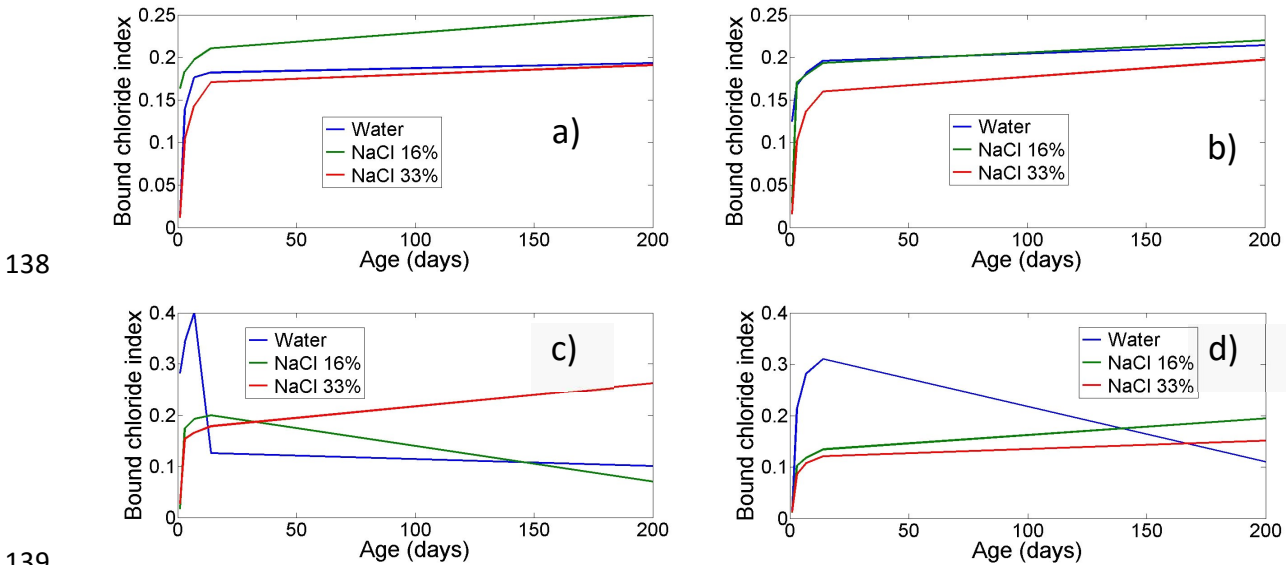
125 The herein proposed ultrasound signal processing to assess the presence of chlorides in the  
 126 specimens was based on the method proposed in our previous work [24], to obtain a bound  
 127 chloride index.

128  $I_c = I - A$  (2)

129 However, in order to distinguish between NaCl concentrations inside the different cement pastes,  
130 a standard deviation of the response of each concentration was obtained with respect to the  
131 response of water-saturated specimens.

#### 132 4. Results and discussion

133 Fig. 2 shows the bound chloride index for the distinct cement pastes using the 50 kHz transverse  
134 wave transducers. Oven drying the samples caused extensive cracking that influenced the  
135 ultrasonic response of the cement pastes. This influence can be seen on the samples containing  
136 water. These specimens present changes that cannot be attributed to the hydration process as the  
137 samples were moist cured for an adequate period of time to ensure a high degree of hydration.

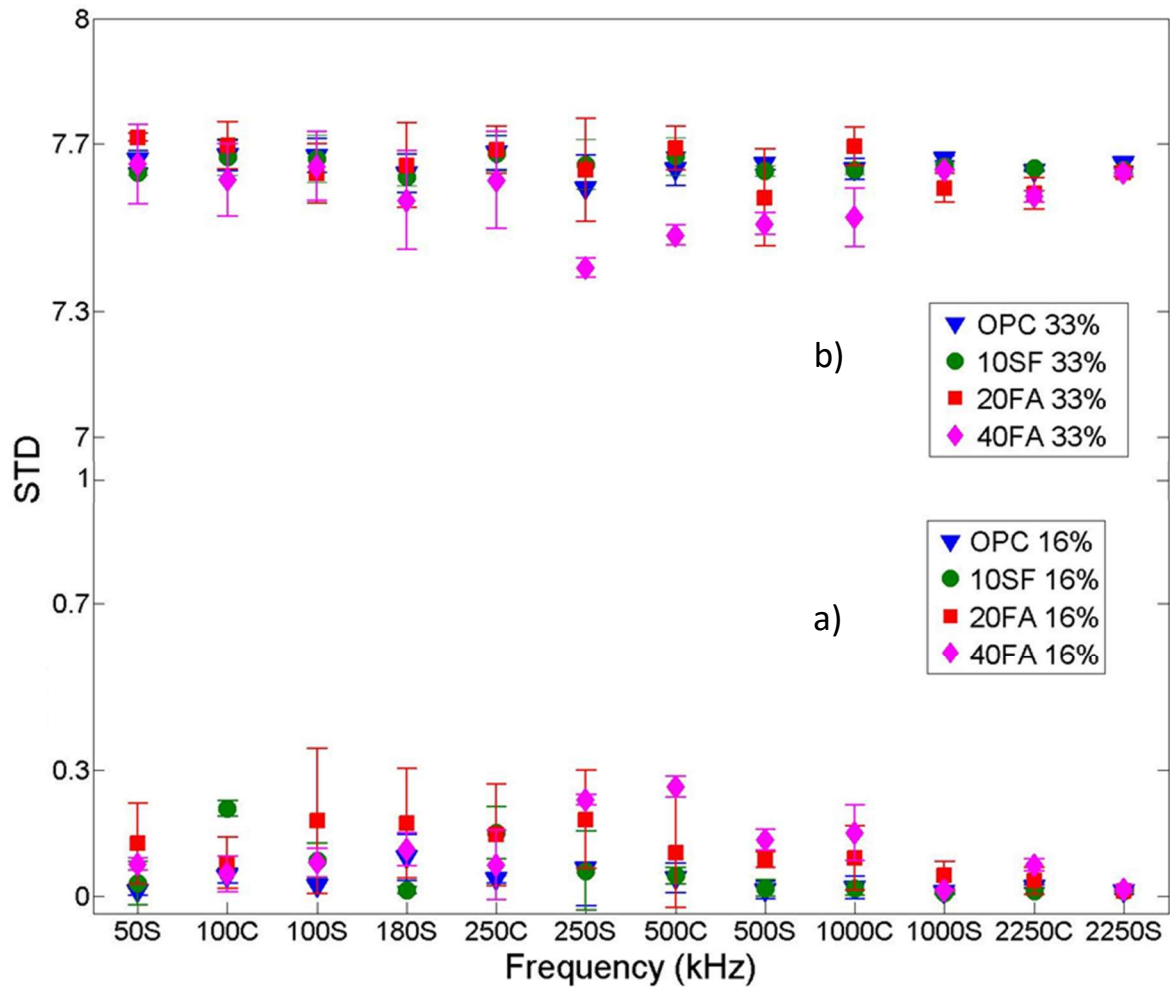


139  
140 Fig. 2. Bound chloride Index obtained for all cement paste specimens containing a) 100 %  
141 Portland cement, b) 10% silica fume, c) 20 % fly ash, and d) 40 % fly ash. Transversal wave  
142 transducers of 50 kHz were employed.

143 These changes may be attributed to the movement of water inside the cement paste after vacuum  
144 saturation. It is possible that water did not completely filled all the gel pores. As a result,  
145 moisture redistribution by capillary forces may occur [25]. In addition, it is not expected that the  
146 penetration of water would significantly change the microstructure of the samples. On the other  
147 hand, chloride chemical binding will produce crystals of Friedel's salt that may affect the  
148 ultrasound response of the samples. It is possible to assess these changes by comparing the index  
149 obtained from the different chloride concentration specimens with the control specimen that  
150 contains only water.

151 The increase in concentration of the NaCl solutions inside the specimens will in turn increase the  
152 viscosity of the pore solution. Higher NaCl concentrations will also increase the amount of  
153 chemically and physically bound chlorides in the cement pastes [26, 27, 28, 29], provided the  
154 binding capacity is not exhausted. This will emphasize the different behavior among the  
155 specimens saturated with different concentrations (0%, 16% and 33% of NaCl). Changes in the  
156 microstructure of the different pastes are due to higher concentration of free chlorides in the pore  
157 solution that reacts with C3A [27]. The latter changes depicted, are reflected in the higher  
158 deviation from the specimens containing water.

159 As can be seen in Fig. 3, it is possible to distinguish the behavior between specimens containing  
160 NaCl solutions with different concentrations, aside of the transducer employed, due to the  
161 difference in average values of standard deviations.



162

163 Fig. 3. Average values of the standard deviation obtained from a) 16 % NaCl solution, and b) 33  
 164 % NaCl solution, assuming as a mean of the process the response of the control specimens  
 165 containing water, for all ultrasound transducers used.

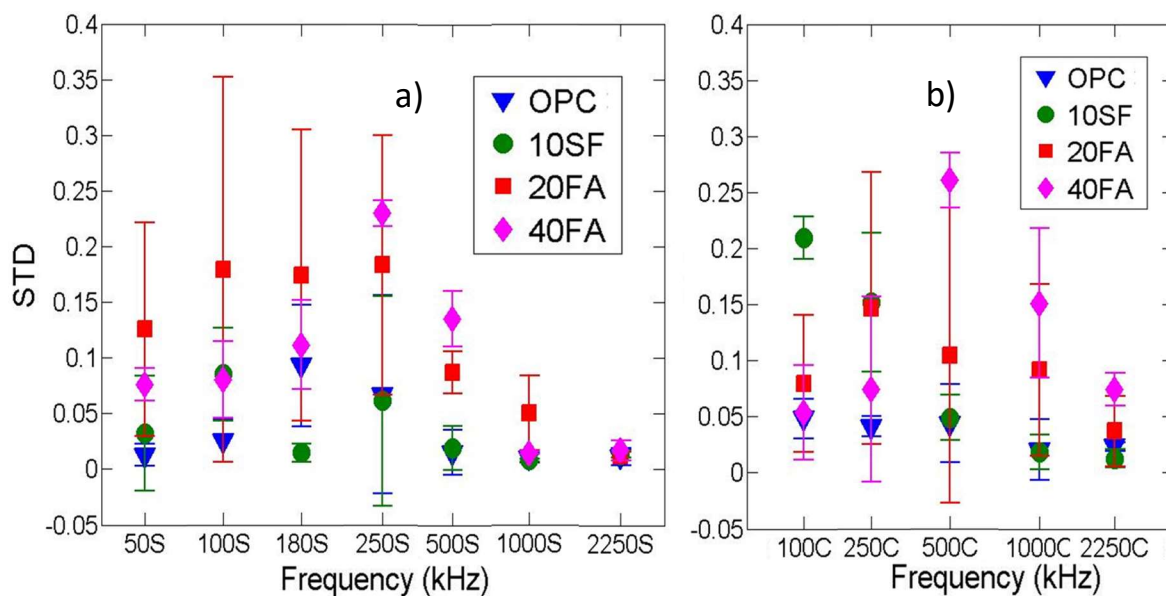
166 Fig. 4 presents the standard deviation of the cement paste specimens with a 16% concentration of  
 167 NaCl versus the control specimens containing water, for all ultrasound transducers. It is observed  
 168 that the standard deviation is very low compared with the 33% concentration specimens.

169 However, as shown in Fig. 4a, the low frequency transducers (50 to 180 kHz) are not sensitive to  
 170 the microstructural changes, unlike the frequencies of 250 and 500 kHz, that are consistent with

171 the chloride binding isotherms shown in Fig. 1. This behavior may be associated with the higher  
 172 sensitivity related to the higher frequencies employed, as well as by the preference of transversal  
 173 waves to travel through the solid phase of the porous medium, modified by the formation of  
 174 Friedel's salt [30].

175 In the case of Fig. 4b, transducers with frequencies from 500 to 2250 kHz show a similar  
 176 behavior compared with the results of the chloride binding isotherms (Fig. 1). This behavior is  
 177 not only due to wavelength but also because of the capability of the longitudinal waves to travel  
 178 through fluid and solid media. Despite of this distinction, the reduction of the concentration of  
 179 NaCl in the pore solution and the binding of the cement paste through the formation of Friedel's  
 180 salt is more difficult to differentiate with this type of waves.

181 As can be seen, the longitudinal wave transducer from 500 kHz and above, had a better  
 182 distinction between the four types of cement paste than the transversal wave transducer. At lower  
 183 frequencies, the results are not consistent.



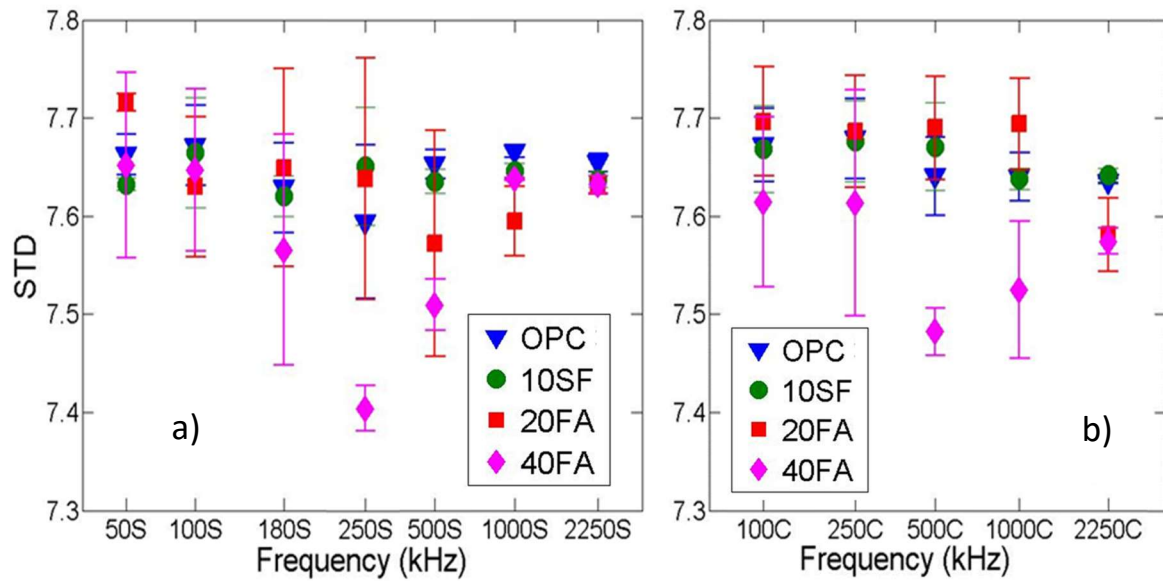
184

185 Fig. 4. Average values of standard deviation for 16 % NaCl solution of all cement paste samples  
186 using a) transversal wave transducers, and b) longitudinal wave transducers.

187 Fig. 5 shows results for the measurements on samples saturated with 33 % NaCl solution  
188 concentration. As can be seen, the standard deviation is higher than that of the 16 % NaCl  
189 concentration. In Fig. 5a the results of the transversal wave transducers show a lower dispersion  
190 of data, in comparison with the last concentration (Fig. 4a). The main feature is that the cement  
191 paste which has the highest binding capacity (40FA) produces a lower average value of the  
192 standard deviation than the other samples. These results may be attributed to the high chloride  
193 concentration, that caused an increase in the pore solution viscosity [31], decreasing the  
194 propagation through the pore structure. Another possible factor is the chemical chloride binding  
195 that obstructs the movement of the pore solution, and subsequently decreasing the ultrasound  
196 signal changes [32]. That may be the main reason why the most reactive cement paste is lower  
197 than the other cement pastes in all transducer frequencies (Fig. 5). However, the preference of  
198 the transversal wave to travel only through the solid phase impact on the high dispersion of the  
199 results. The results of the 250 kHz transversal wave transducer show an acceptable  
200 differentiation among the different cement pastes

201 Results of the longitudinal wave transducers are shown in Fig. 5b. As can be seen, the behavior  
202 is very similar to the transversal wave transducer. However, since the longitudinal waves are able  
203 to travel through fluid and solid media, the resulting data dispersion is lower than that of the  
204 transversal wave transducers. The higher frequencies ranging from 500 and 1000 kHz present a  
205 better differentiation among cement pastes. Unlike the results for 16% NaCl, in this case even for  
206 the lower frequencies it is possible to distinguish the different cement pastes





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208 Fig. 5. Average values of standard deviation for 33% NaCl solutions for all cement paste  
 209 specimens, a) transversal wave transducer, and b) longitudinal wave transducer.

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## 221 **5. Conclusions**

- 222 1. It is possible to distinguish between the chloride concentrations of the solutions  
223 employed for saturating the samples (0 %, 16 % and 33 %) considering as reference  
224 the deionized water saturated specimens. This applies for all four-different cement  
225 pastes.
- 226 2. The results obtained with the proposed methodology, for the 16 % NaCl concentration  
227 it is consistent with the chloride binding isotherms, better identified with the 250 and  
228 500 kHz transversal wave transducer and longitudinal wave transducer with  
229 frequencies of 500 kHz and higher, despite the extensive cracking of the specimens  
230 due to oven drying.
- 231 3. Results of the ultrasound signals analysis obtained with the methodology proposed, for  
232 the specimens saturated with 33 % NaCl concentration, are explained by the high  
233 concentration of chloride in the pore solution that increase the amount of chloride  
234 bound in the cement pastes, consuming the aluminates. This can be better appreciated  
235 in 250 kHz transversal wave transducer and for 500 and 1000 kHz longitudinal wave  
236 transducer.

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